

Synthesis and Electronic Properties of Iron-Sulfur Clusters Relevant to Metalloenzymes

Kazuyuki Tatsumi and Yasuhiro Ohki

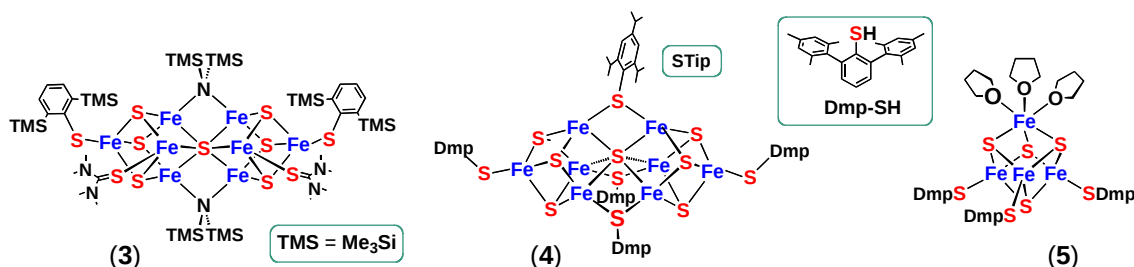
Research Center for Materials Science, and Department of Chemistry, Graduate School
of Science, Nagoya University, Furo-cho, Chikusa-ku, Nagoya 464-8602, Japan

E-mail; i45100a@nucc.cc.nagoya-u.ac.jp

The iron-sulfur proteins in nature are represented by those consisting of [4Fe-4S], [3Fe-4S], and [2Fe-2S] clusters. In most of the cases, these clusters are thermodynamically stable, and the core geometries are constructed by spontaneous self-assembly reactions in polar aprotic solvents. We have recently found a new method to synthesize Fe/S clusters, which utilizes the bis-amide complex of iron, $\text{Fe}\{\text{N}(\text{SiMe}_3)_2\}_2$, as the precursor. This new route has allowed us to build Fe/S clusters in non-polar solvents, and resulted in a series of unprecedented cluster structures, some of which resemble closely the active sites of nitrogenase and hydrogenase.

For instance, the unusual [8Fe-7S] inorganic core structure of the reduced form of P-cluster (P^{N}) of nitrogenase was self-assembled by the reaction of $\text{Fe}\{\text{N}(\text{SiMe}_3)_2\}_2$, tetramethylthiourea (*tmtu*), 2, 4, 6-triisopropylbenzenethiol(*HS-tip*), and S_8 in the ratio of 8(Fe):3(*tmtu*):12(*HS-tip*):7(S), which gave rise to $[\{\text{N}(\text{TMS})_2\}\{\text{SC}(\text{NMe}_2)_2\}\text{Fe}_4\text{S}_3]_2(\mu_6\text{-S})\{\mu\text{-N}(\text{TMS})_2\}_2$ ($\text{TMS} = \text{SiMe}_3$)(**1**). Interestingly, the cluster **1** consists of irons with a 6Fe(II)+2Fe(III) oxidation state as manifested by the Mössbauer study, while P^{N} has been thought to carry 8 Fe(II) ions. Although cluster **1** degrades to a Fe_4S_4 cubane cluster under the presence of thiols, as the P-cluster does so, it was possible to isolate $(\text{NEt}_4)_2[\{\text{N}(\text{TMS})_2\}(\text{SAr})\text{Fe}_4\text{S}_3]_2(\mu_6\text{-S})\{\mu\text{-N}(\text{TMS})_2\}_2$ (**2**) ($\text{Ar} = 4\text{-}^t\text{Bu-C}_6\text{H}_4$, 2-SiMe₃-C₆H₄ etc.) and $[(\text{SAr})\{\text{SC}(\text{NMe}_2)_2\}\text{Fe}_4\text{S}_3]_2(\mu_6\text{-S})\{\mu\text{-N}(\text{TMS})_2\}_2$ ($\text{Ar} = \{2,6\text{-(SiMe}_3)_2\text{-C}_6\text{H}_4\}$) (**3**) from the reactions of **1** with corresponding thiolates and thiol. We have isolated yet another type of [8Fe-7S] cluster, $[(\text{SDmp})\text{Fe}_4\text{S}_3]_2(\mu_6\text{-S})(\mu\text{-SDmp})_2(\mu\text{-STip})$ (**4**), from the reaction of the preformed di-iron complex $[\text{Fe}(\text{Stip})]_2(\mu\text{-SDmp})_2$ with S_8 , which is topologically analogous to FeMo-co of nitrogenase and has an intriguing 5Fe(II)+3Fe(III) oxidation state with a doublet ground state.

We have also synthesized an unusual [4Fe-4S] cluster, $\text{Fe}_4\text{S}_4(\text{SDmp})_3(\text{THF})_3$ (**5**) from $\text{Fe}_4\text{S}_4\{\text{N}(\text{TMS})_2\}_4$ in toluene. Cluster **5** was further converted to $\text{Fe}_4\text{S}_4(\text{SDmp})_3\text{(Imd)}$ (**6**) (Imd = tetramethyl imidazole), which serves as the first model of the histidine-bound unique [4Fe-4S] cluster found in hydrogenase.



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