

分子性固体中の伝導性 π 電子

Conducting π electrons in molecular solids

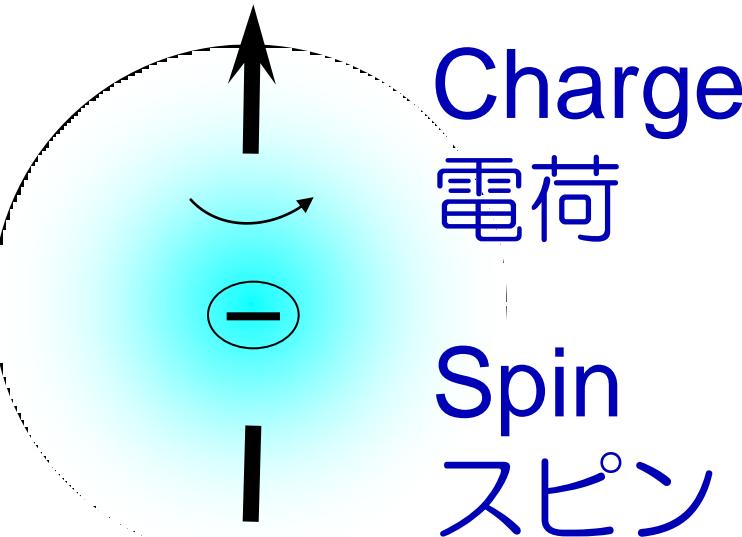
加藤礼三

Reizo KATO

Fundamental Questions in Solid State Science

What are metals ?

What are magnets ?



The *electron* is the lead
of solid state science.

Chinese Periodic Table

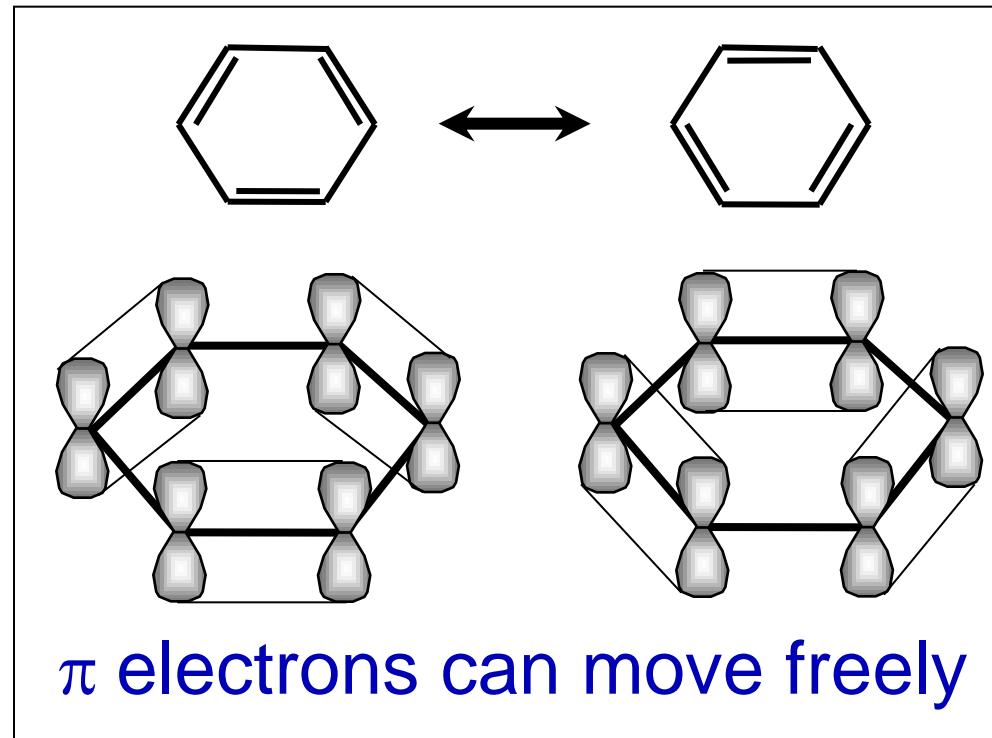
¹ H																				² He
氢																				氦
₃ Li	₄ Be																			₁₀ Ne
锂	铍																			氖
₁₁ Na	₁₂ Mg																			₁₈ Ar
钠	镁																			氩
₁₉ K	₂₀ Ca	₂₁ Sc	₂₂ Ti	₂₃ V	₂₄ Cr	₂₅ Mn	₂₆ Fe	₂₇ Co	₂₈ Ni	₂₉ Cu	₃₀ Zn	₃₁ Ga	₃₂ Ge	₃₃ As	₃₄ Se	₃₅ Br	₃₆ Kr			
钾	钙	钪	钛	钒	铬	锰	铁	钴	镍	铜	锌	镓	锗	砷	硒	溴	氪			
₃₇ Rb	₃₈ Sr	₃₉ Y	₄₀ Zr	₄₁ Nb	₄₂ Mo	₄₃ Tc	₄₄ Ru	₄₅ Rh	₄₆ Pd	₄₇ Ag	₄₈ Cd	₄₉ In	₅₀ Sn	₅₁ Sb	₅₂ Te	₅₃ I	₅₄ Xe			
铷	锶	钇	锆	铌	钼	锝	钌	铑	钯	钯	镉	铟	锡	锑	碲	碘	氙			
₅₅ Cs	₅₆ Ba	lanthanoid	₇₂ Hf	₇₃ Ta	₇₄ W	₇₅ Re	₇₆ Os	₇₇ Ir	₇₈ Pt	₇₉ Au	₈₀ Hg	₈₁ Tl	₈₂ Pb	₈₃ Bi	₈₄ Po	₈₅ At	₈₆ Rn			
铯	钡		铪	钽	钨	铼	锇	铱	铂	金	汞	铊	铅	铋	钋	钋	氡			
₈₇ Fr	₈₈ Ra	actinoid	₁₀₄ Rf	₁₀₅ Db	₁₀₆ Sg	₁₀₇ Bh	₁₀₈ Hs	₁₀₉ Mt	₁₁₀ Ds	₁₁₁ Rg	₁₁₂ Cp	₁₁₃	₁₁₄ Uuq		₁₁₆ Uuh					
钫	镭		𬬻	𬭊	𬭳	𬭛	𬭛	𬭛												

lanthanoid	₅₇ La	₅₈ Ce	₅₉ Pr	₆₀ Nd	₆₁ Pm	₆₂ Sm	₆₃ Eu	₆₄ Gd	₆₅ Tb	₆₆ Dy	₆₇ Ho	₆₈ Er	₆₉ Tm	₇₀ Yb	₇₁ Lu
	镧	铈	镨	钕	钷	钐	铕	钆	铽	镝	钬	铒	铥	镱	镥
actinoid	₈₉ Ac	₉₀ Th	₉₁ Pa	₉₂ U	₉₃ Np	₉₄ Pu	₉₅ Am	₉₆ Cm	₉₇ Bk	₉₈ Cf	₉₉ Es	₁₀₀ Fm	₁₀₁ Md	₁₀₂ No	₁₀₃ Lr
	锕	钍	镤	铀	镎	钚	镅	锔	锫	锫	锿	镄	钔	锘	铹

Metals from Non-metal Elements

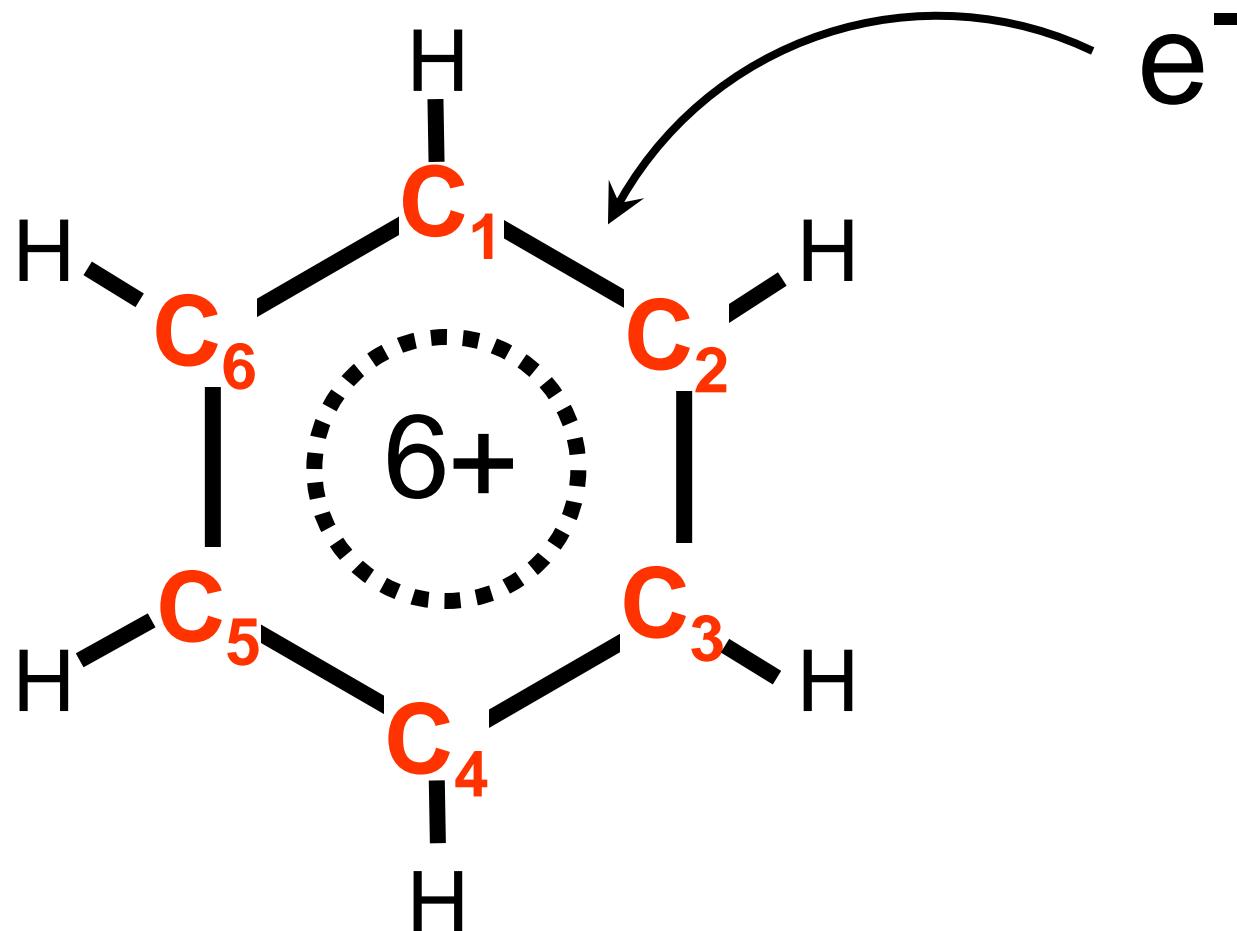
4

Metals have conducting electrons which can move in a crystal freely and carry electricity and heat.



A π electron in benzene

5



Fundamental Principles (I)

The superposition of probability amplitudes (確率振幅の重ね合わせ)

$$\psi(t) = \sum_i C_i(t) \phi_i$$

ϕ_i : Base state

$C_i(t)$: Probability amplitude

Any state can be represented as a linear combination of a set of **base states** with suitable coefficients (**probability amplitudes**).

系の任意の状態は「ベクトル」のように「基本状態」の線形結合で表される。

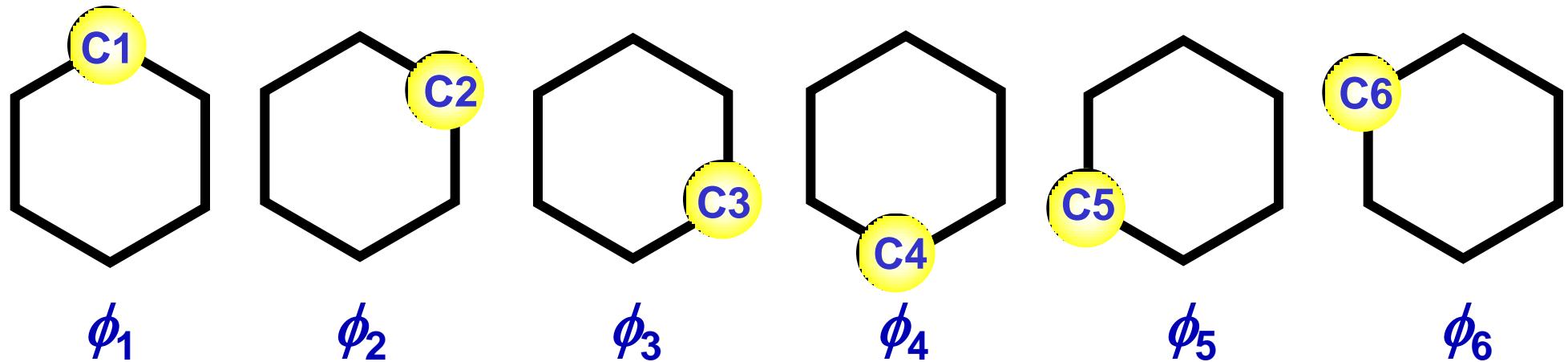
The probability P_i that the system is in the base state ϕ_i is the square of the absolute value of the probability amplitude C_i .

系が基本状態 ϕ_i にある確率 P_i は、確率振幅 C_i の絶対値の 2 乗。 $P_i = |C_i(t)|^2$

The probability amplitude is a **complex number**.

確率振幅は**複素数**。

Base states for a π electron in benzene 7



$$\psi(t) = \sum_i^6 C_i(t) \phi_i$$

Fundamental Principles (II)

8

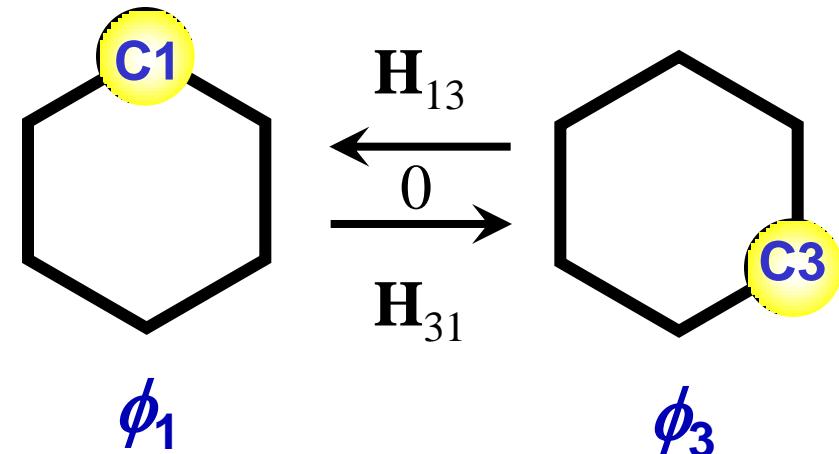
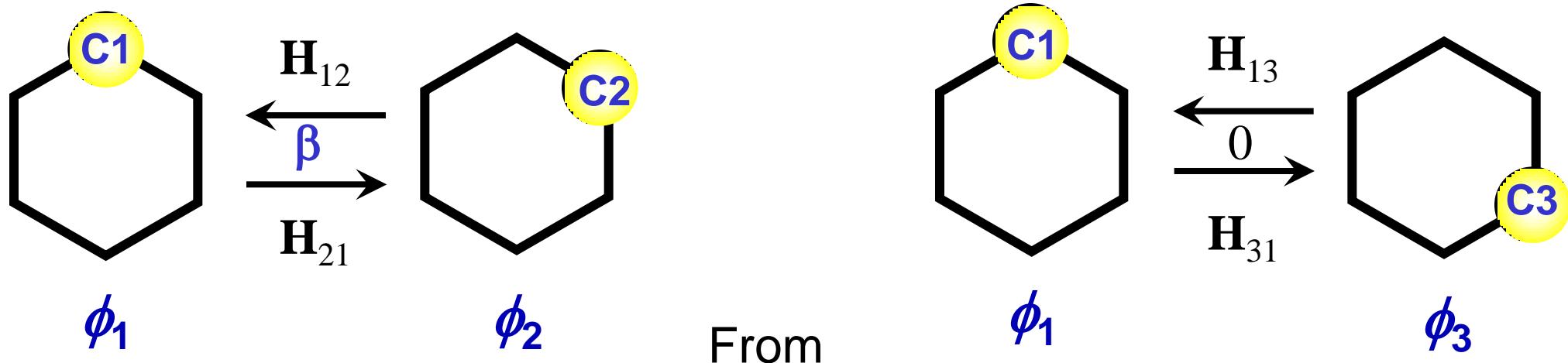
How the amplitudes vary with time

$$i\hbar \frac{dC_i(t)}{dt} = \sum_j \mathbf{H}_{ij} C_j(t)$$

\mathbf{H}_{ij} : Hamiltonian matrix
(ハミルトニアン行列)

– $(i/\hbar)\mathbf{H}_{ij}$ is the amplitude that the base state j will be converted into the base state i during the time dt .
(基本状態 j が、時間 dt の間に、基本状態 i へ移る確率振幅)

H_{ij} in benzene ?

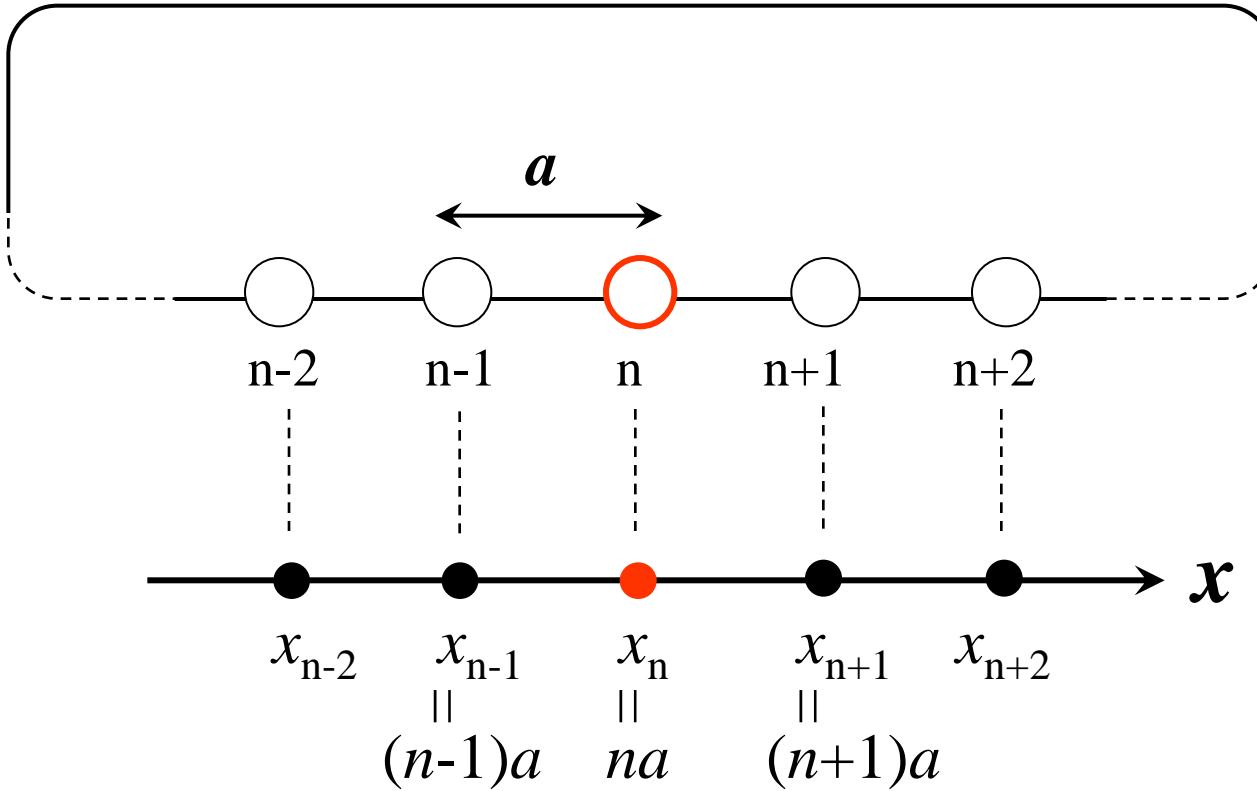


To

H_{ij}	ϕ_1	ϕ_2	ϕ_3	ϕ_4	ϕ_5	ϕ_6
ϕ_1	α	β	0	0	0	β
ϕ_2	β	α	β	0	0	0
ϕ_3	0	β	α	β	0	0
ϕ_4	0	0	β	α	β	0
ϕ_5	0	0	0	β	α	β
ϕ_6	β	0	0	0	β	α

N -membered Ring

10



$$i\hbar \frac{dC_n(t)}{dt} = \alpha C_n(t) + \beta C_{n+1}(t) + \beta C_{n-1}(t)$$

Fundamental Principles (III)

Stationary states (定常状態)

States in which all the amplitudes have the same time dependence $e^{-\frac{i}{\hbar}Et}$

確率振幅 $C_i(t)$ がすべて同じ時間依存性 $e^{-\frac{i}{\hbar}Et}$ を持つ状態

Stationary states are states of definite energy E .
この時、系は確定したエネルギー値 E を持つ。

$$C_i(t) = A e^{-\frac{i}{\hbar}Et}$$

The wave traveling in the x -direction with the wave length λ

$$e^{ikx} \cdot e^{-i\omega t} = e^{ikx} \cdot e^{-\frac{i}{\hbar}Et} \quad k = \frac{2\pi}{\lambda} : \text{wave number (波数)}$$

$$\hbar\omega = \frac{h\omega}{2\pi} = h\nu = E \quad (\leftarrow \text{light quanta: Einstein})$$

N-membered Ring

$$i\hbar \frac{dC_n(t)}{dt} = \alpha C_n(t) + \beta C_{n+1}(t) + \beta C_{n-1}(t)$$

In the stationary state, $C_n(t) = A(x_n)e^{-\frac{i}{\hbar}Et}$

$$EA(x_n) = \alpha A(x_n) + \beta A(x_{n+1}) + \beta A(x_{n-1})$$

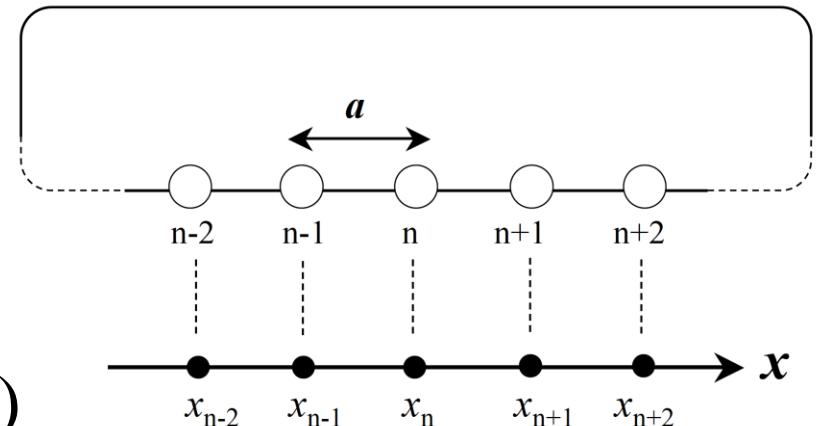
$$= \alpha A(x_n) + \beta A(x_n + a) + \beta A(x_n - a)$$

Let's take as a trial solution $A(x_n) = A_0 e^{ikx_n}$ (the space dependence of the amplitudes)

$$E(k) = \alpha + \beta e^{ika} + \beta e^{-ka} = \boxed{\alpha + 2\beta \cos ka}$$

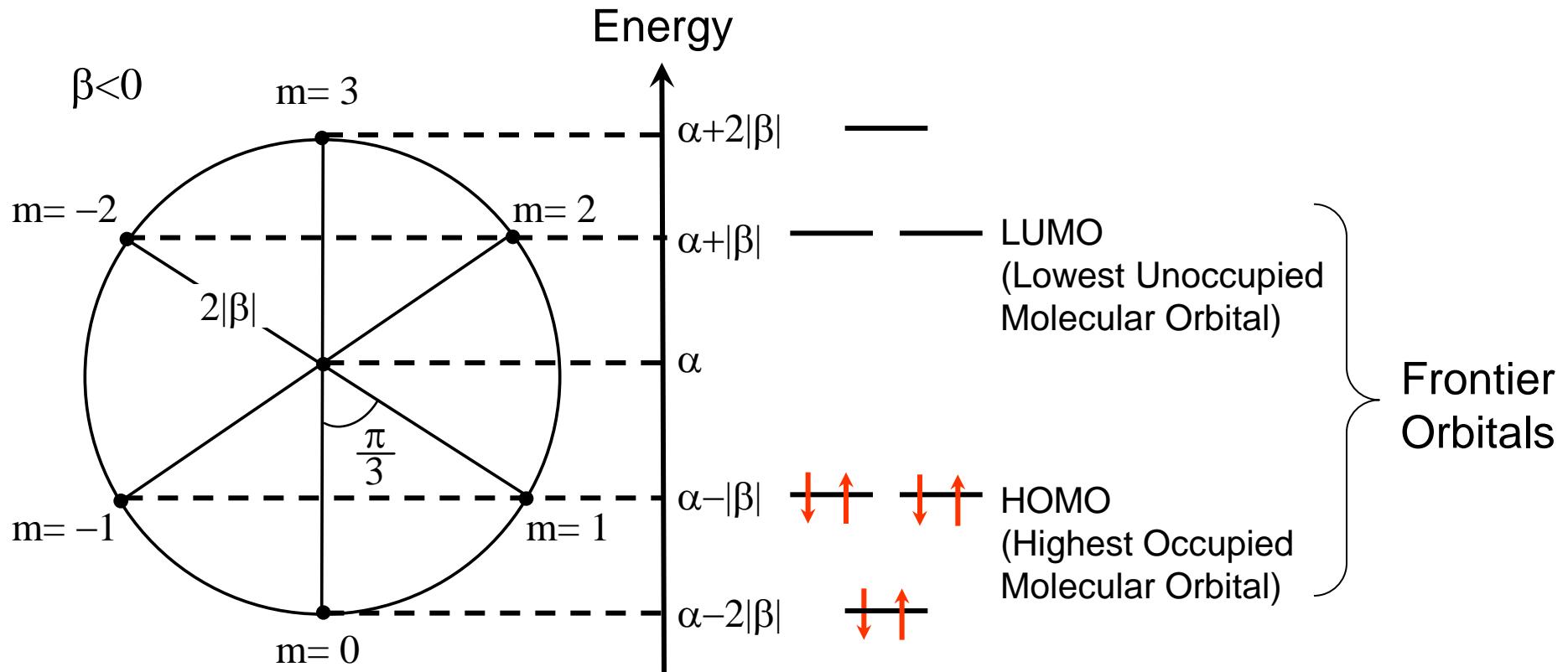
$$A(x_{n+N}) = A(x_n + Na) = e^{ik(x_n + Na)} = A(x_n) = e^{ikx_n}$$

$$\therefore e^{ikNa} = 1 \quad i.e. \quad \boxed{k = \frac{2\pi}{a} \cdot \frac{1}{N} \cdot m \quad \text{for integral } m}$$



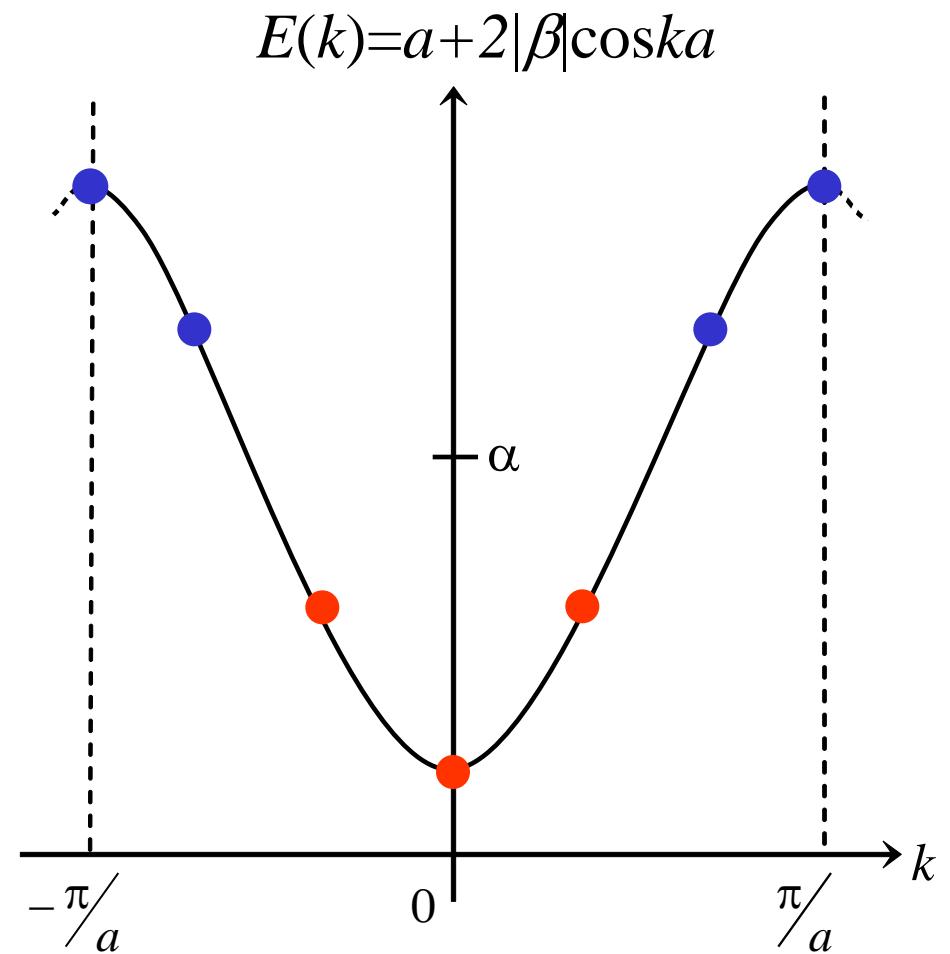
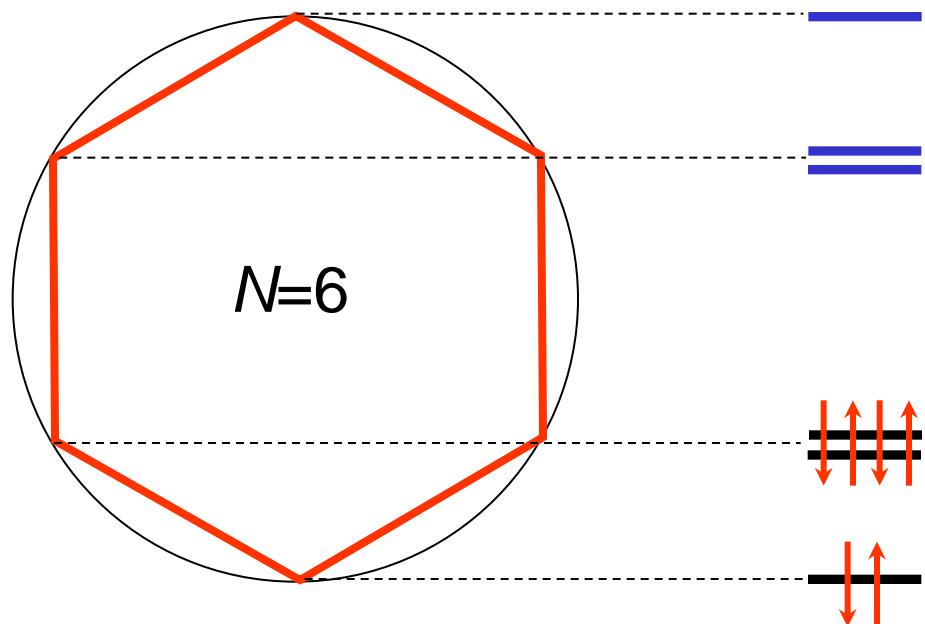
Energy levels in benzene

$$\left. \begin{aligned} E(k) &= \alpha + 2\beta \cos ka \\ k &= \frac{2\pi}{a} \cdot \frac{1}{N} \cdot m = \frac{2\pi}{a} \cdot \frac{1}{6} \cdot m \end{aligned} \right\} E = \alpha + 2\beta \cos\left(\frac{\pi}{3}m\right)$$



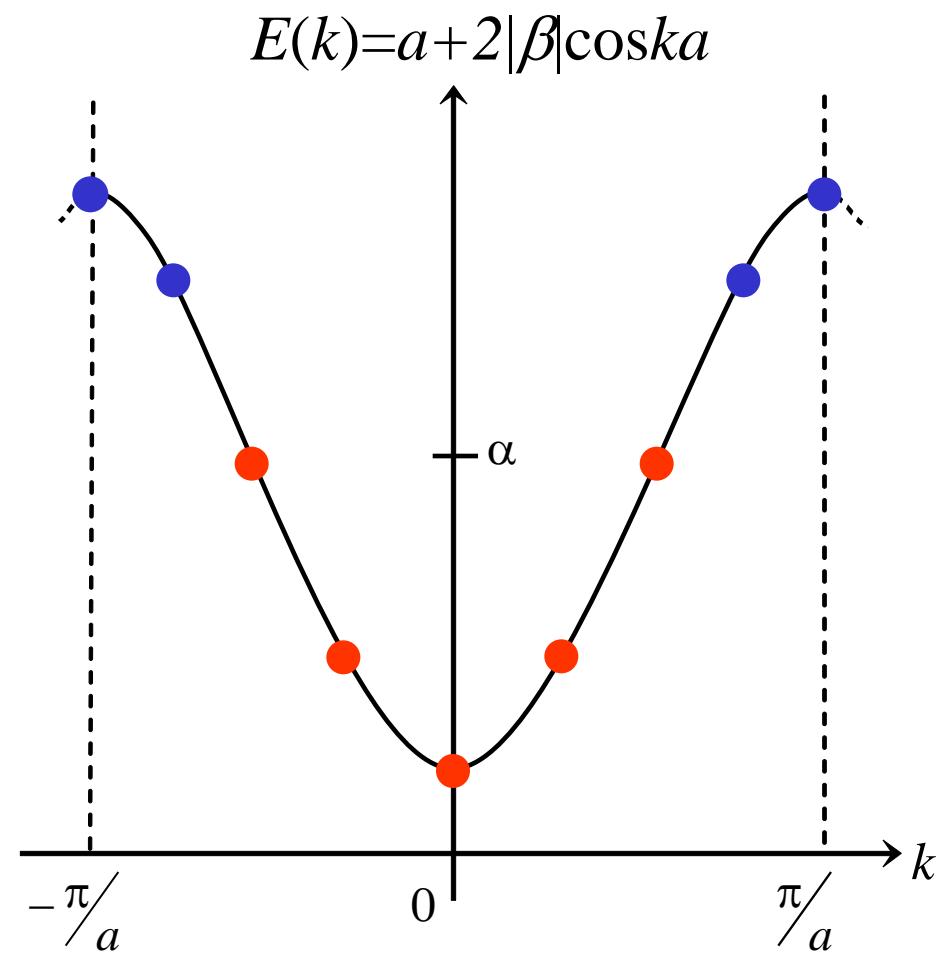
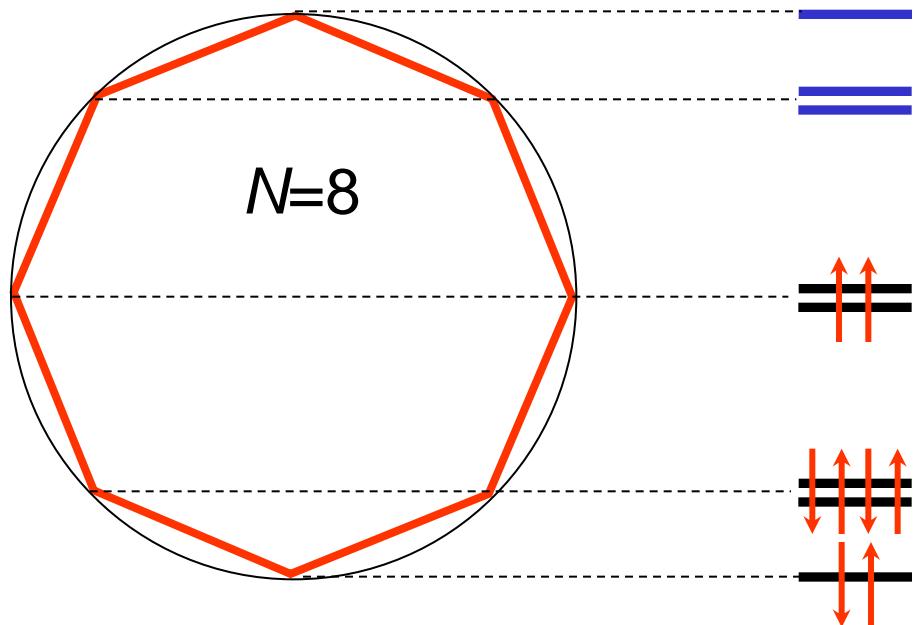
Energy levels ($N = 6$)

14



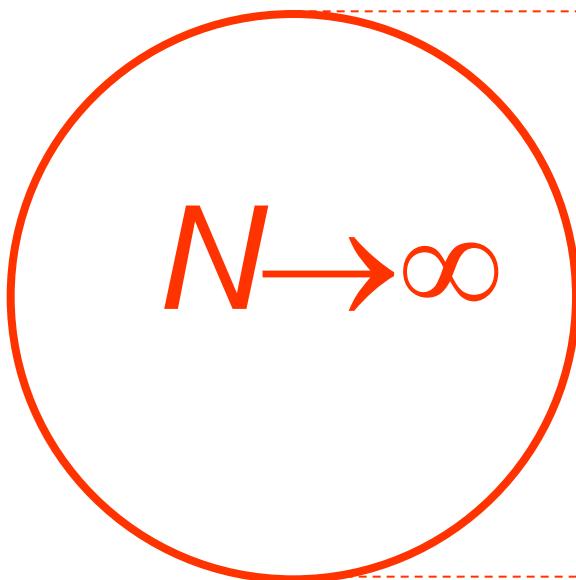
Energy levels ($N = 8$)

15

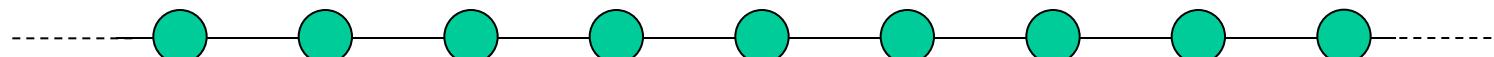
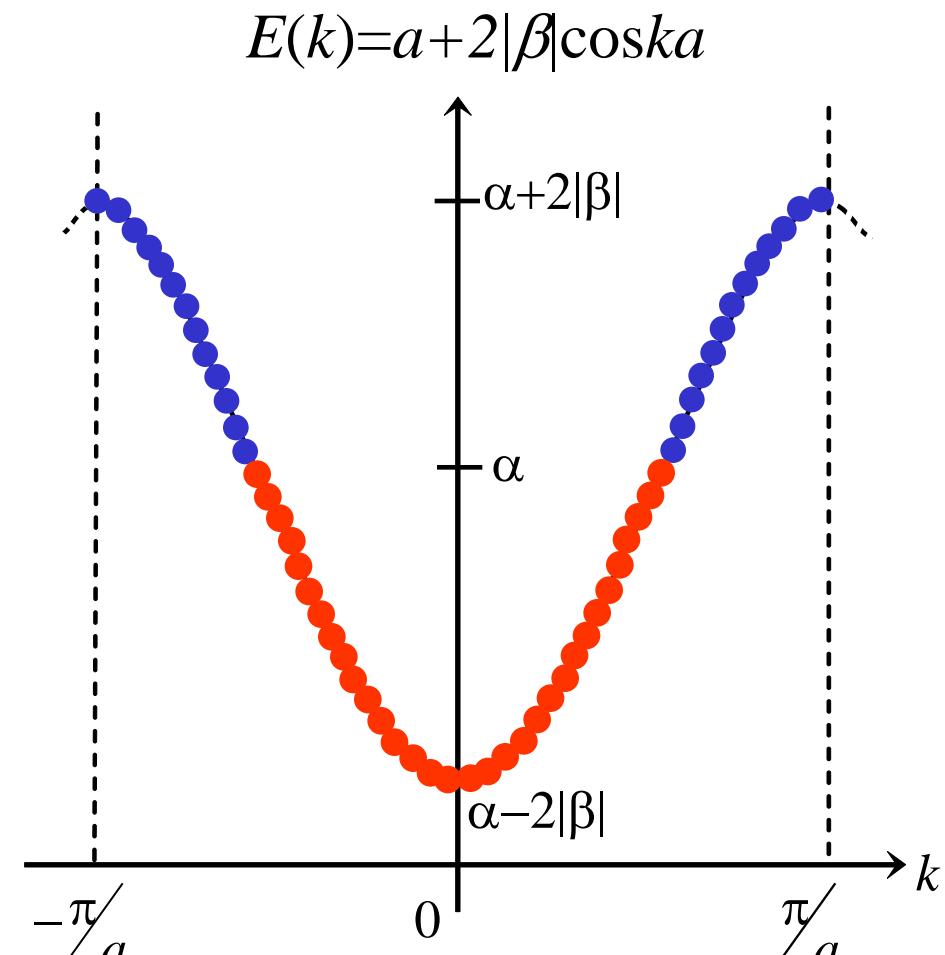
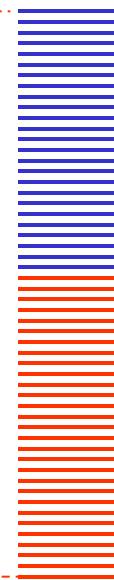


$N \rightarrow \infty$: Energy band

16



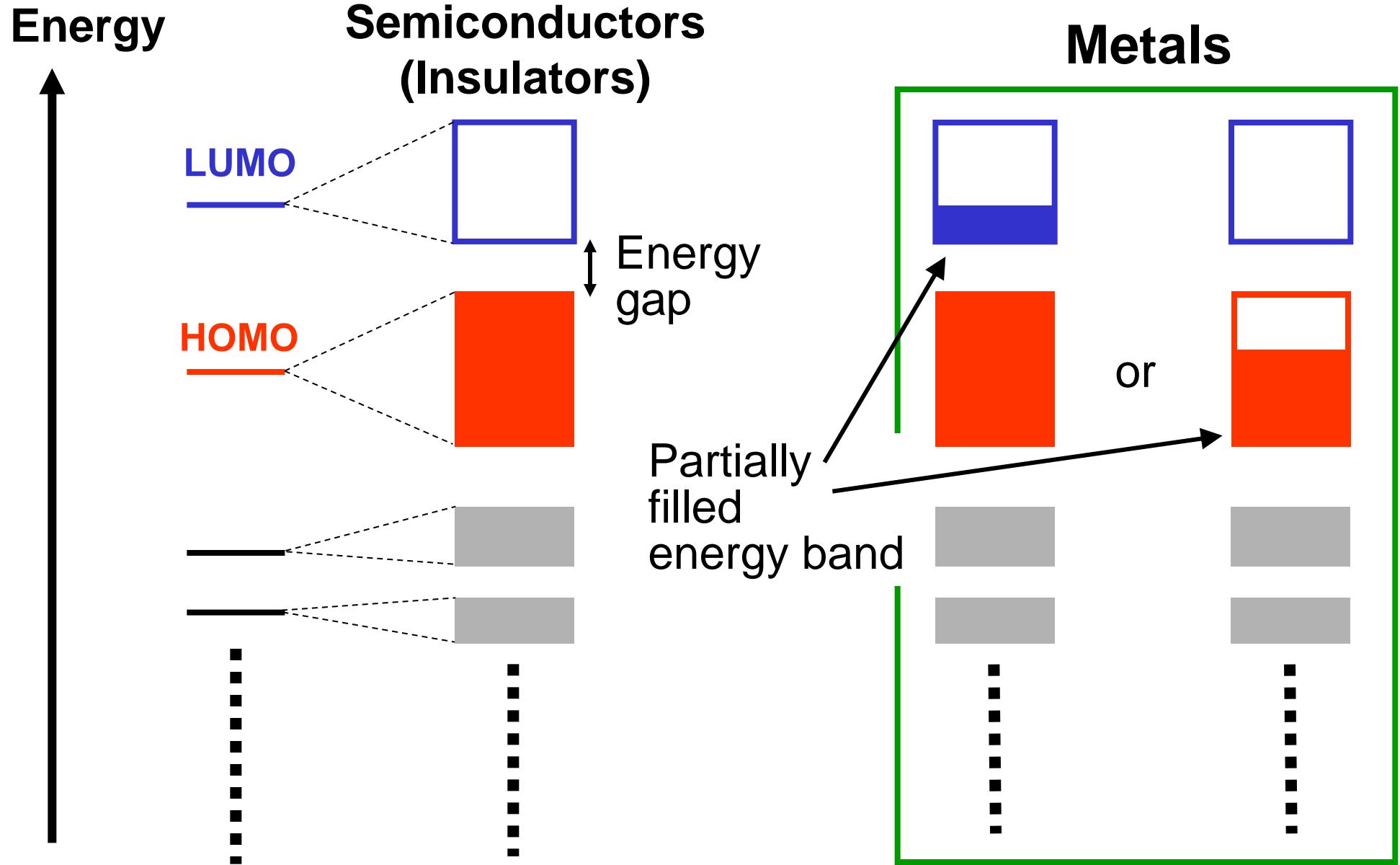
Energy
band



One-dimensional (uniform) lattice

What are metals ?

17



The Nobel Prize in Chemistry (2000)

18

Conductive polymers



Polyacetylene film



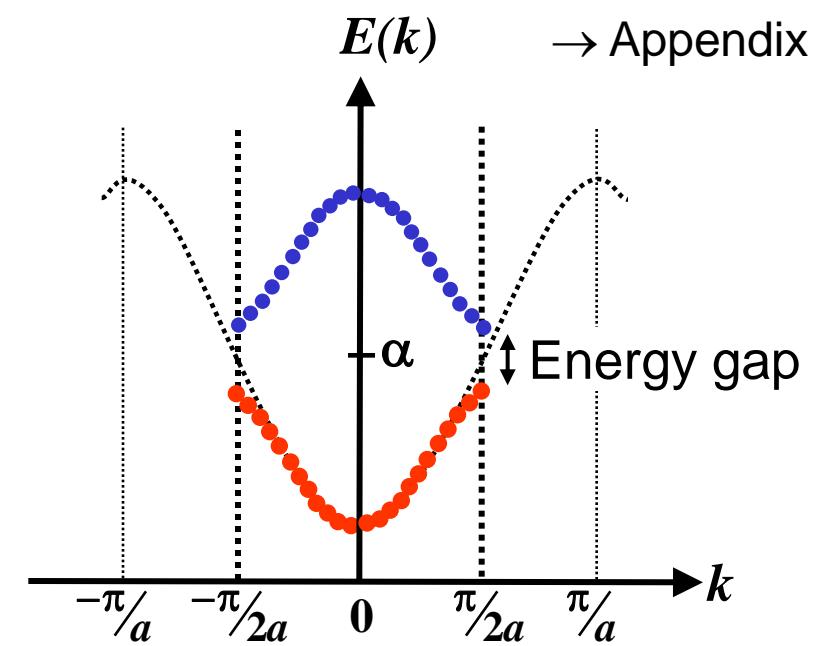
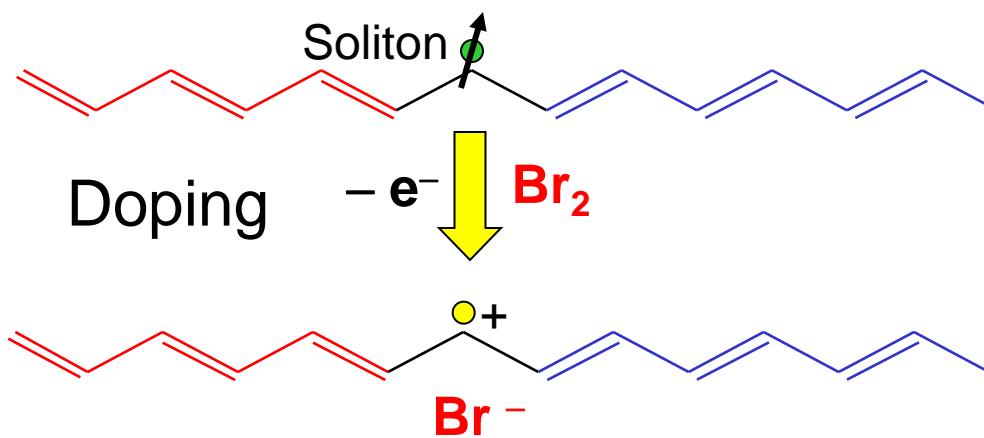
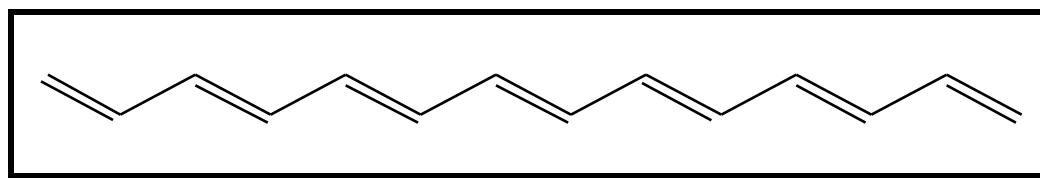
Hideki Shirakawa



Alan J. Heeger



Alan G. MacDiarmid



The Nobel Prize in Physics (2010)

19



Graphite

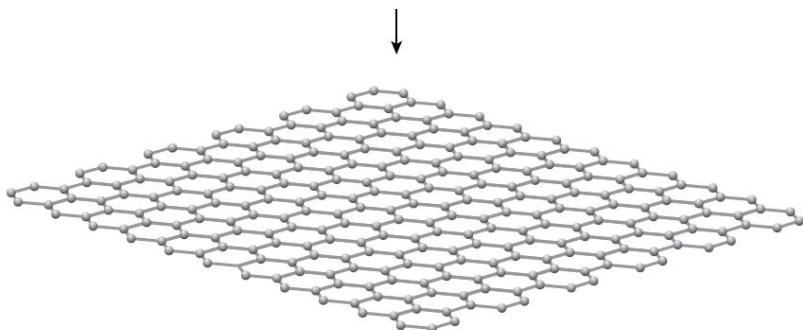
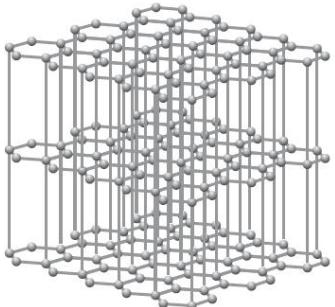


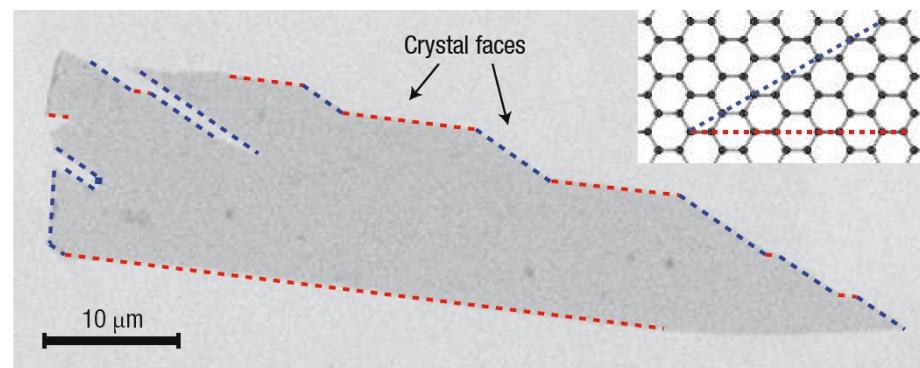
Photo: Sergeom, Wikimedia Commons

Andre Geim



Photo: University of Manchester, UK

Konstantin
Novoselov



Graphene (Two-dimensional crystal)

Description of crystals

-Lattice and repeated unit-

A (Bravais) **lattice** is an infinite array of discrete points (離散的な点)

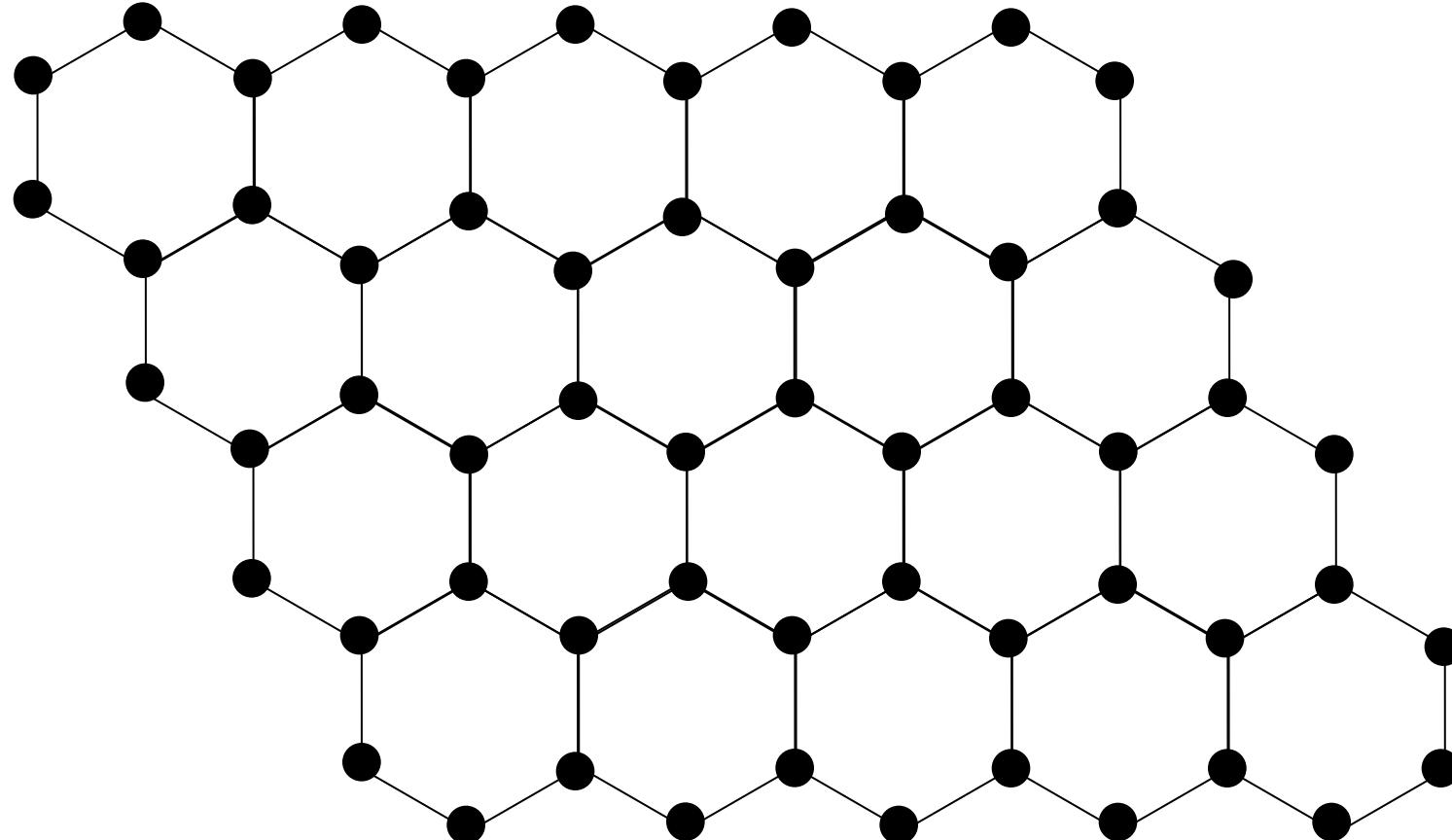
- (a) with an arrangement and orientation that appears exactly the same, from whichever of the points the array is viewed.
どこの点から配列を眺めてみても、その配列と配向が全く同一
- (b) with position vectors (位置ベクトル) \mathbf{r} of the form
$$\mathbf{r} = n_1\mathbf{a} + n_2\mathbf{b} + n_3\mathbf{c}$$
 for integral n_1 , n_2 , and n_3
 $\mathbf{a}, \mathbf{b}, \mathbf{c}$: Primitive vectors

A Bravais lattice specifies the periodic array in which the repeated units are arranged.

ブラベー格子は繰り返し単位の周期配列を特徴付ける。

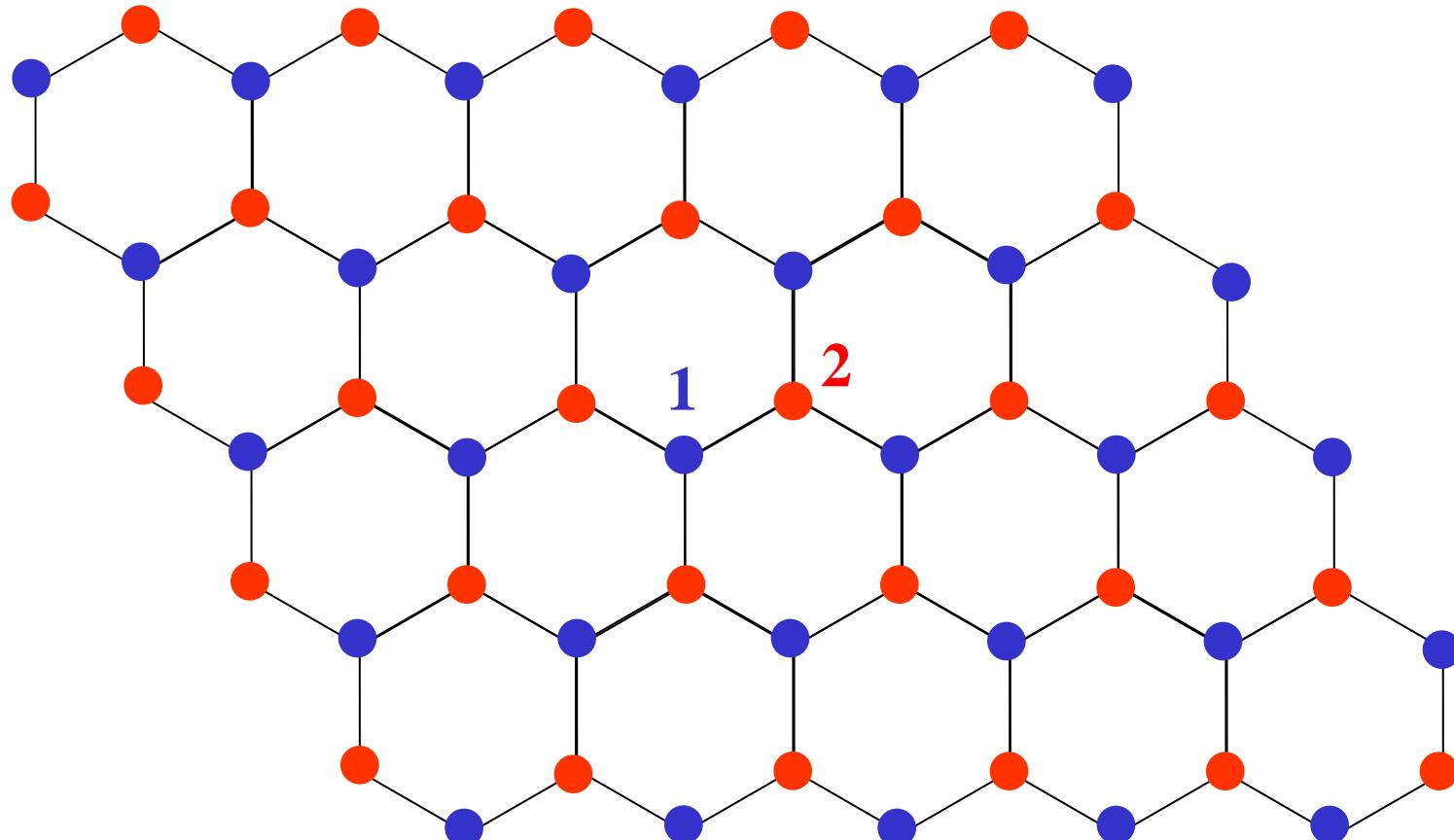
Is this a (Bravais) lattice ?

21



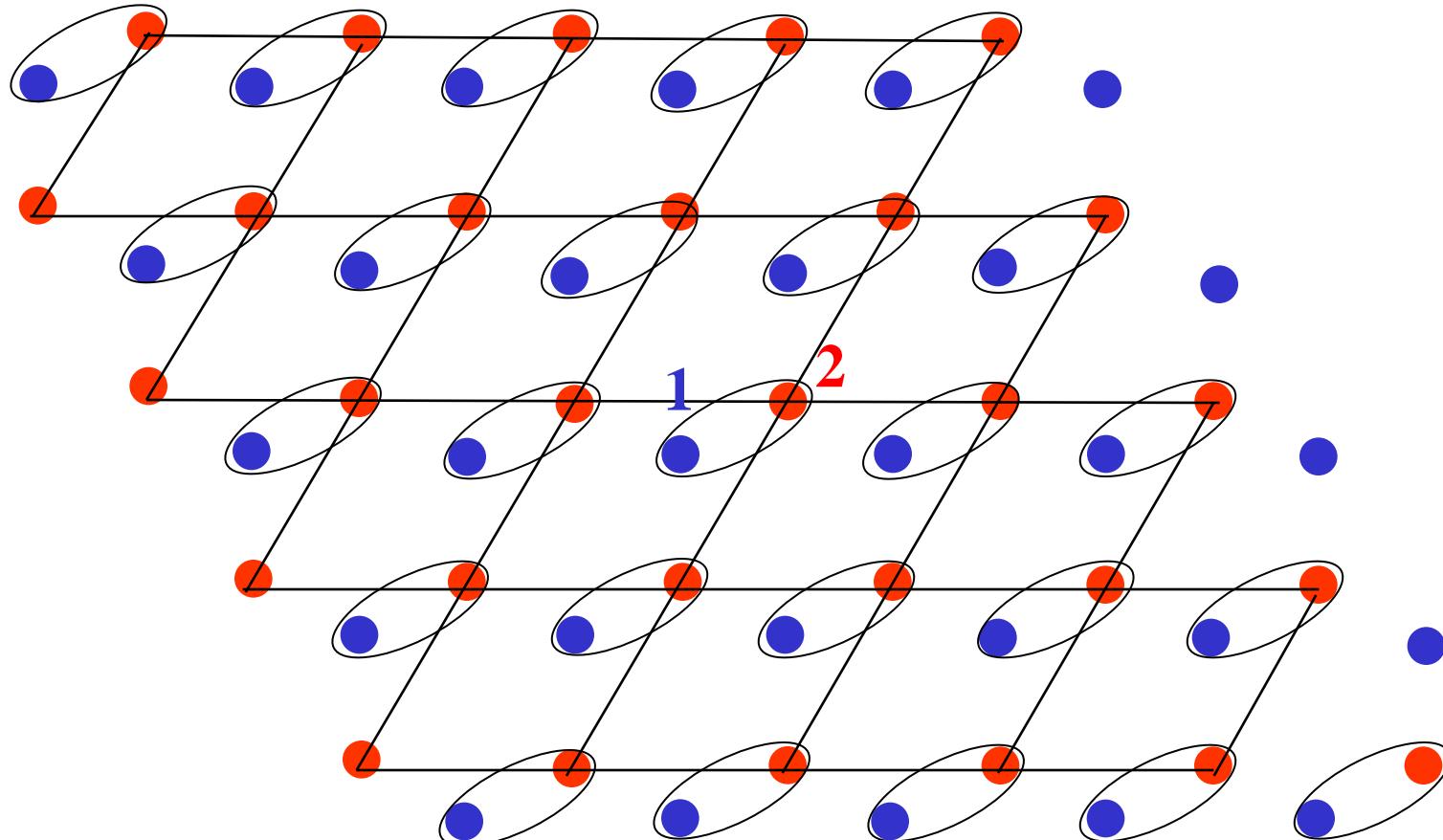
What is a repeated unit ?

22



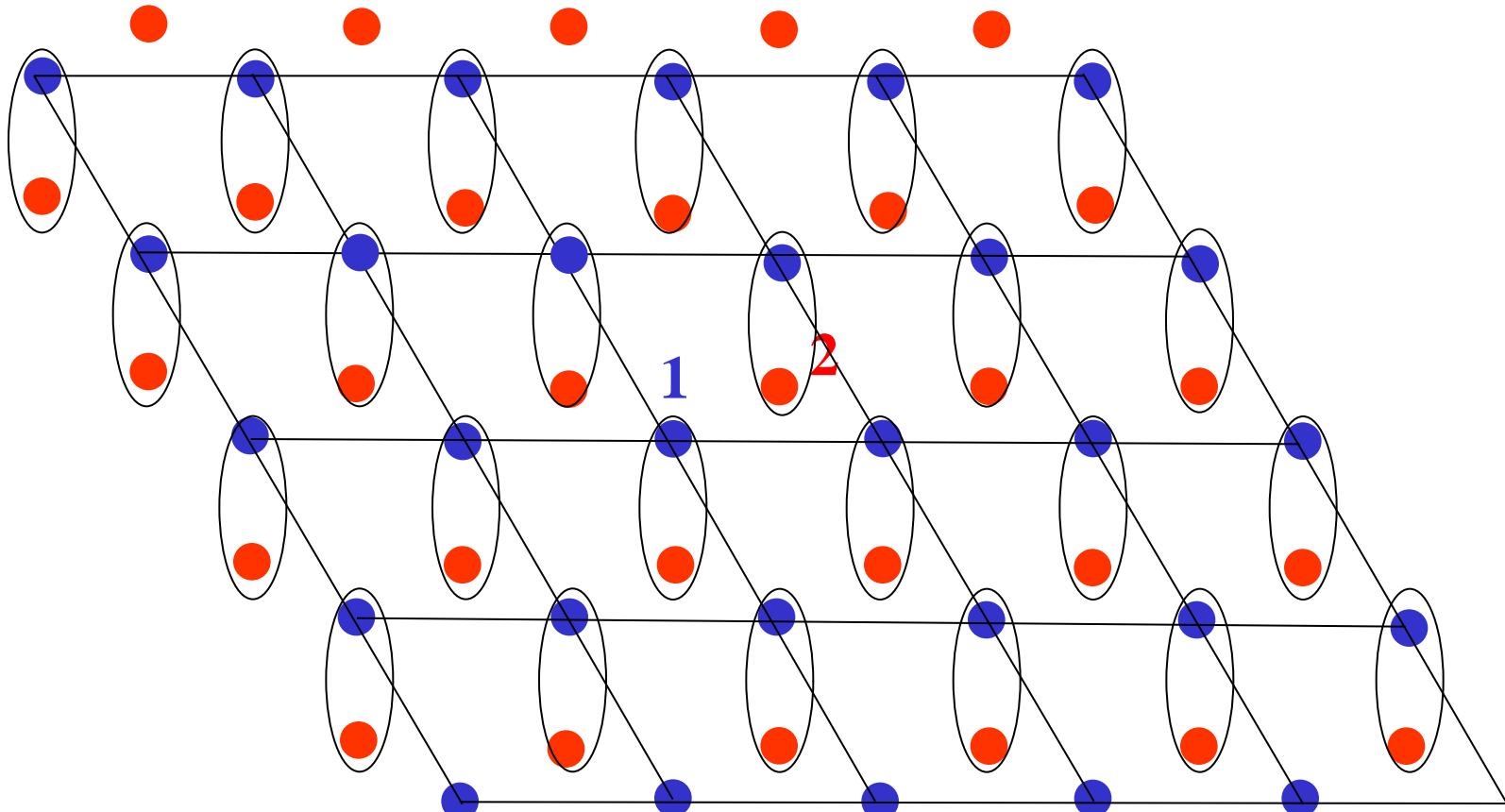
How to describe Graphene crystal

23



Another description

24

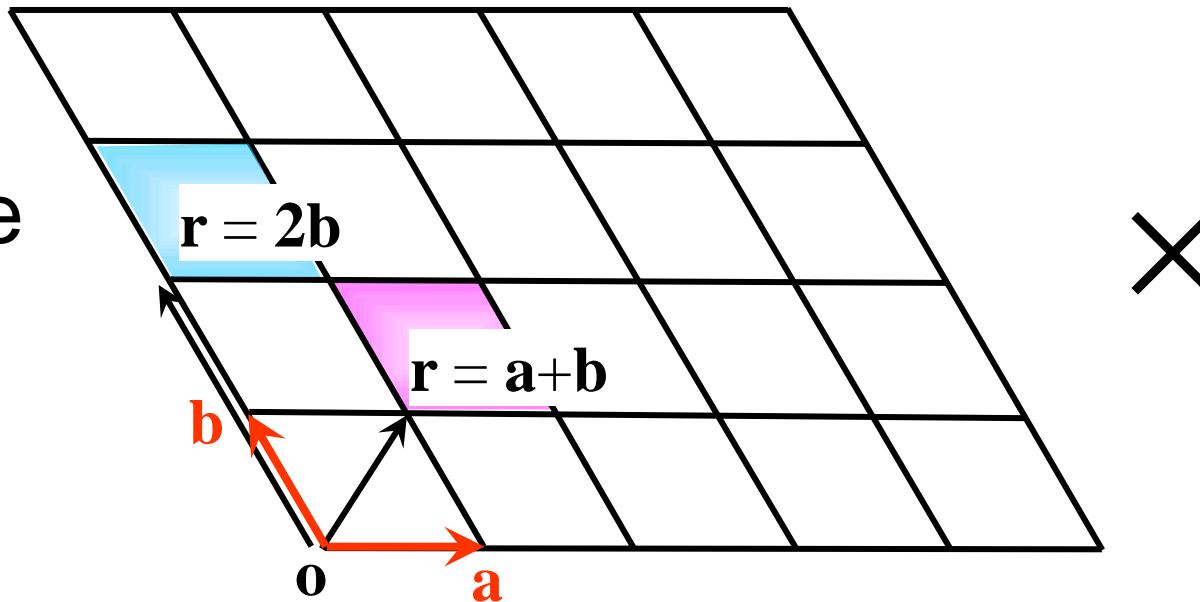


Graphene crystal

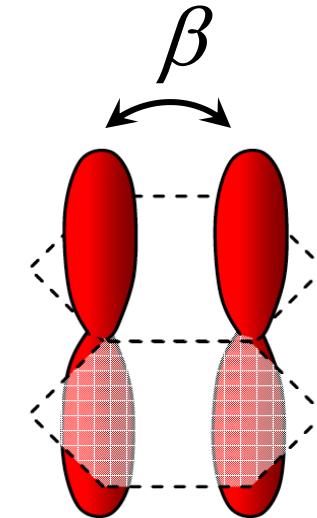
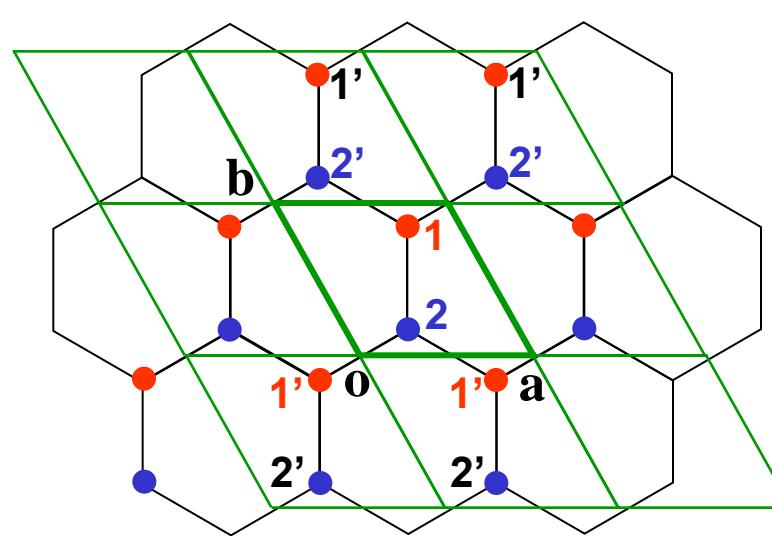
Graphene
crystal =

Lattice

Repeated
unit



How the amplitudes vary with time in graphene ? 26



$$\left\{ \begin{array}{l} i\hbar \frac{dC_{1,\mathbf{0}}}{dt} = \beta C_{2,\mathbf{0}} + \beta C_{2,\mathbf{b}} + \beta C_{2,(\mathbf{a}+\mathbf{b})} \\ i\hbar \frac{dC_{2,\mathbf{0}}}{dt} = \beta C_{1,\mathbf{0}} + \beta C_{1,-\mathbf{b}} + \beta C_{1,-(\mathbf{a}+\mathbf{b})} \end{array} \right.$$

(The origin of the energy level is chosen so that α is zero.)

Energy band for graphene

In the stationary state, $C_{i,\mathbf{r}}(t) = A_i(\mathbf{r}_n) \cdot e^{-\frac{i}{\hbar}Et}$

$$\begin{cases} EA_1(\mathbf{o}) = \beta A_2(\mathbf{o}) + \beta A_2(\mathbf{b}) + \beta A_2(\mathbf{a} + \mathbf{b}) \\ EA_2(\mathbf{o}) = \beta A_1(\mathbf{o}) + \beta A_1(-\mathbf{b}) + \beta A_1(-\mathbf{a} - \mathbf{b}) \end{cases}$$

Let's take as a trial solution $A_i(\mathbf{r}_n) = A_i e^{i\mathbf{k} \cdot \mathbf{r}_n} (= A_i e^{i(k_x x + k_y y)})$

$$\begin{cases} -EA_1 + (\beta + \beta e^{i\mathbf{k} \cdot \mathbf{b}} + \beta e^{i\mathbf{k} \cdot (\mathbf{a} + \mathbf{b})})A_2 = 0 \\ (\beta + \beta e^{-i\mathbf{k} \cdot \mathbf{b}} + \beta e^{-i\mathbf{k} \cdot (\mathbf{a} + \mathbf{b})})A_1 - EA_2 = 0 \end{cases}$$

未知数 A_1, A_2 に対する
連立 1 次方程式

This is a set of linear algebraic equations for the unknowns A_1 and A_2 , and there is a solution if the determinant of the coefficients of A_1 and A_2 is zero.

$$\begin{vmatrix} -E & \beta + \beta e^{i\mathbf{k} \cdot \mathbf{b}} + \beta e^{i\mathbf{k} \cdot (\mathbf{a} + \mathbf{b})} \\ \beta + \beta e^{-i\mathbf{k} \cdot \mathbf{b}} + \beta e^{-i\mathbf{k} \cdot (\mathbf{a} + \mathbf{b})} & -E \end{vmatrix} = 0$$

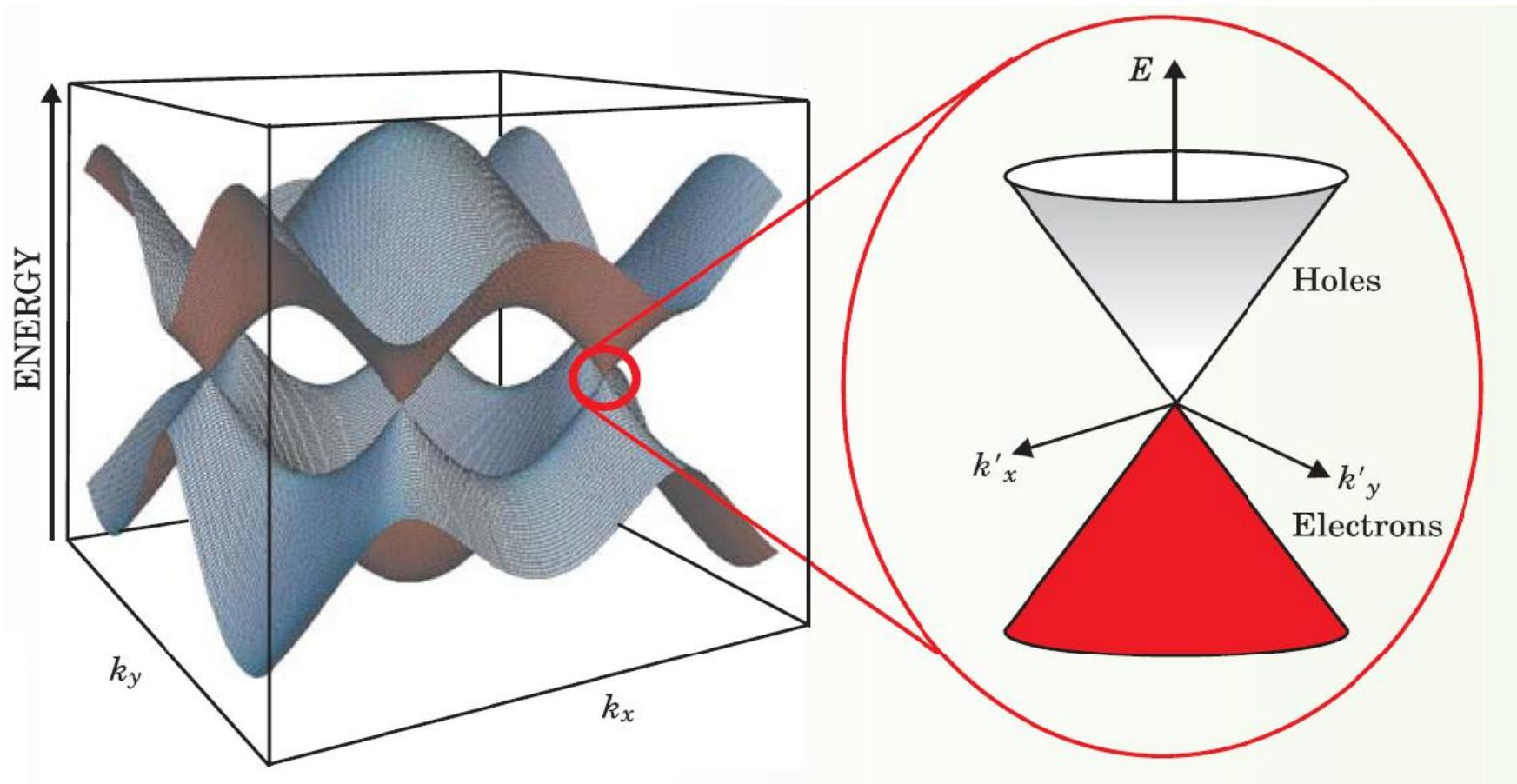
cf. 1D chain

$$C(t) = A(x_n) \cdot e^{-\frac{i}{\hbar}Et}$$

$$A(x_n) = A_o e^{ikx_n}$$

Dirac cone

$$E(\mathbf{k}) = \pm |\beta| \sqrt{3 + 2\cos\mathbf{k} \cdot \mathbf{a} + 2\cos\mathbf{k} \cdot \mathbf{b} + 2\cos\mathbf{k} \cdot (\mathbf{a} + \mathbf{b})}$$

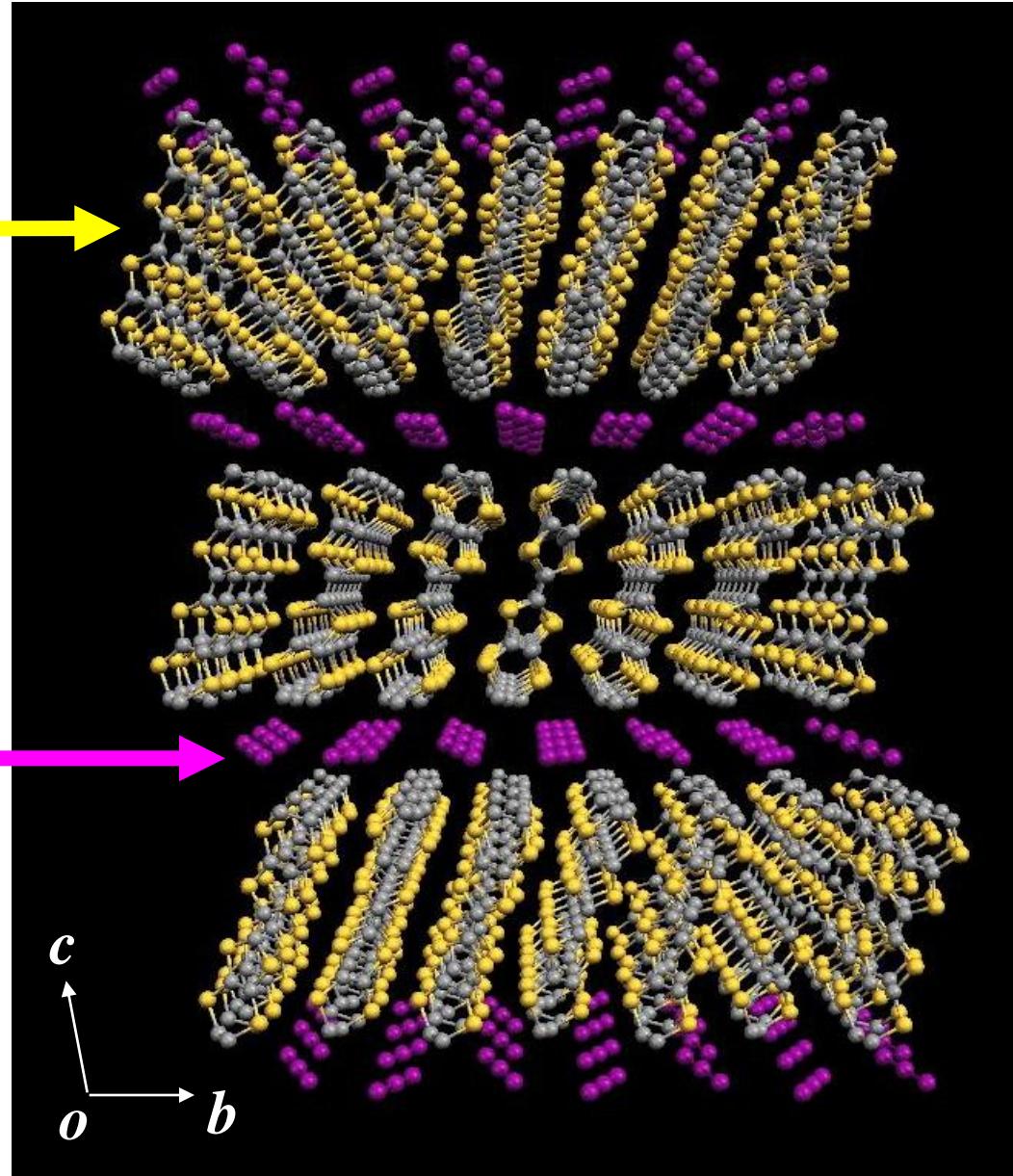
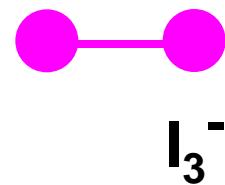
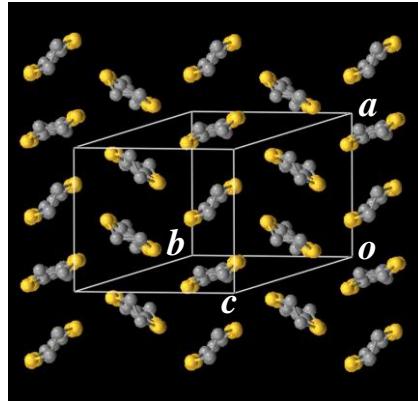
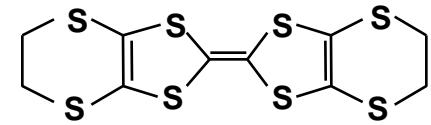


Zero-gap conductor

Bulk zero-gap conductor: α -(ET)₂I₃

29

BEDT-TTF (ET)

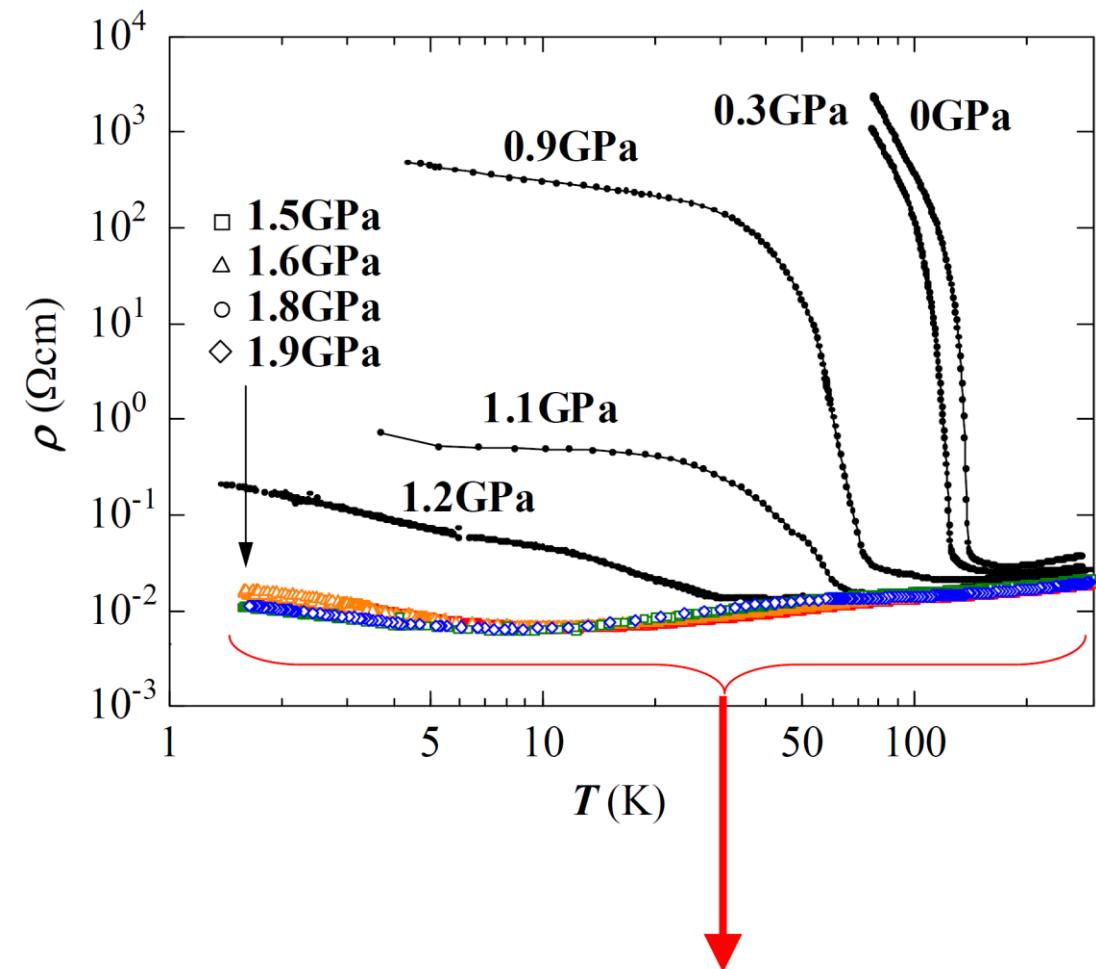


Dr. N. Tajime
(RIKEN)

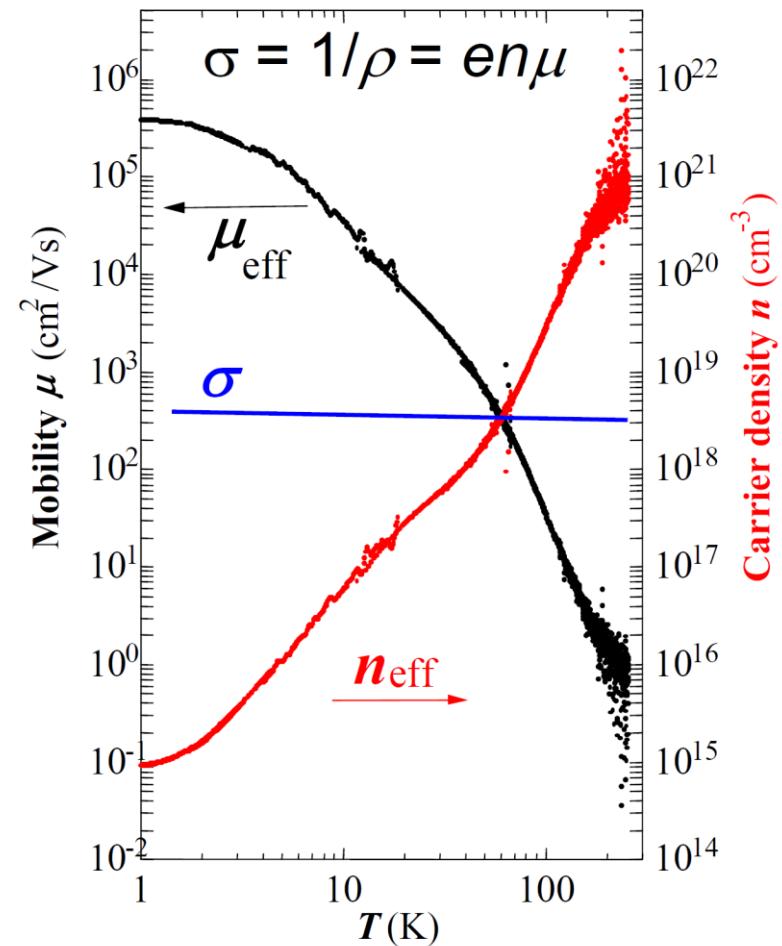
- 1) *J. Phys. Soc. Jpn.*, **69**, 543 (2000).
 - 2) *Europhys. Lett.*, **80**, 47002 (2007).
 - 3) *Phys. Rev. Lett.*, **102**, 176403 (2009).

$\alpha\text{-}(\text{ET})_2\text{I}_3$: Resistivity under hydrostatic pressure

30

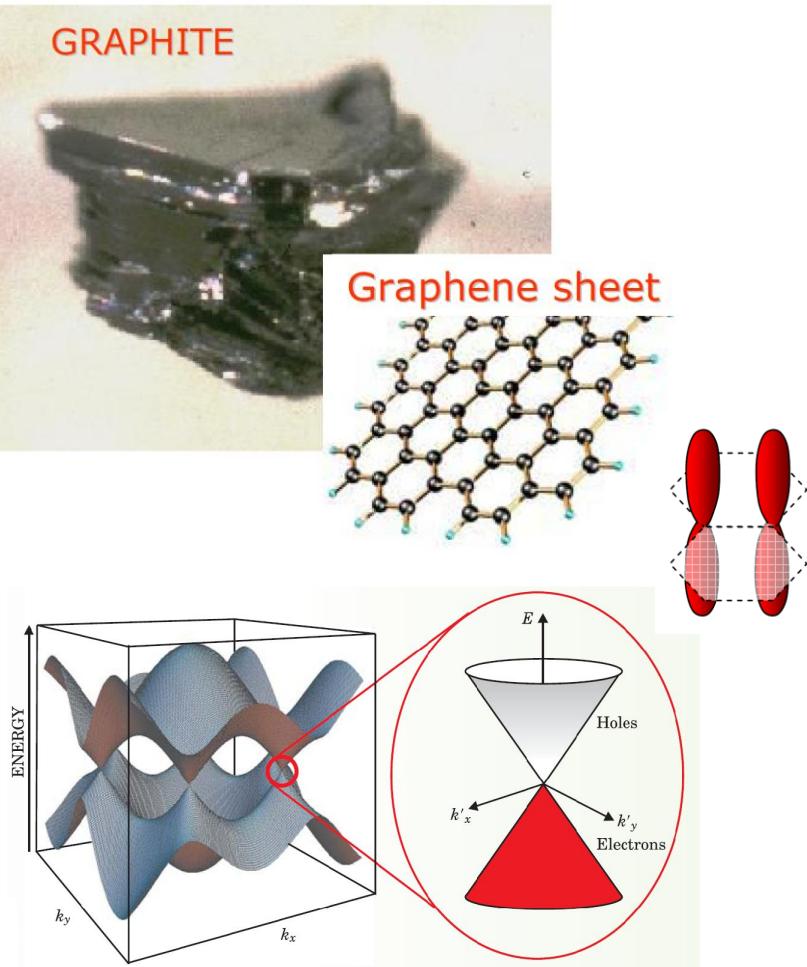


T -independent resistivity



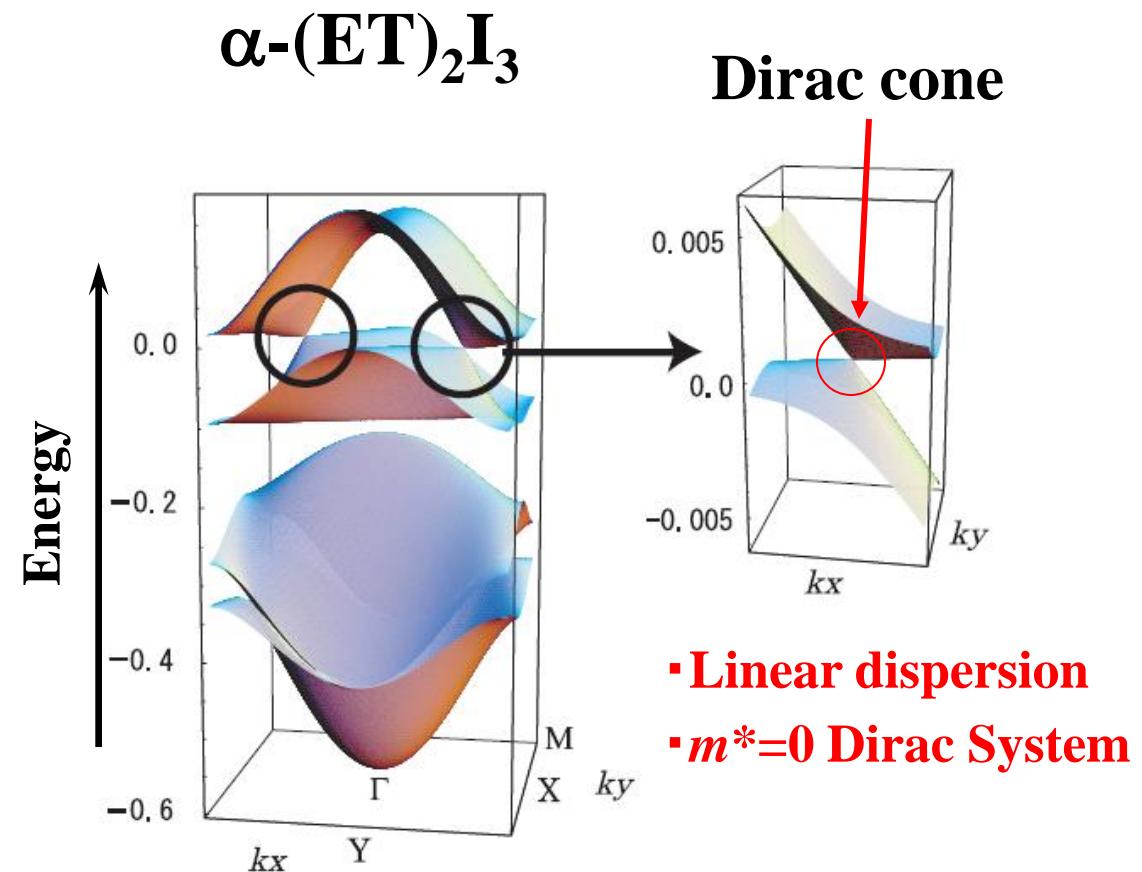
Carrier density and mobility change by about six orders of magnitude, in a manner so that the effects just cancel out.

Graphene



K. S. Novoselov, A. K. Geim et al.,
Nature, **438**, 197 (2005).

Y. Zhang et al., *Nature*, **438**, 201 (2005).

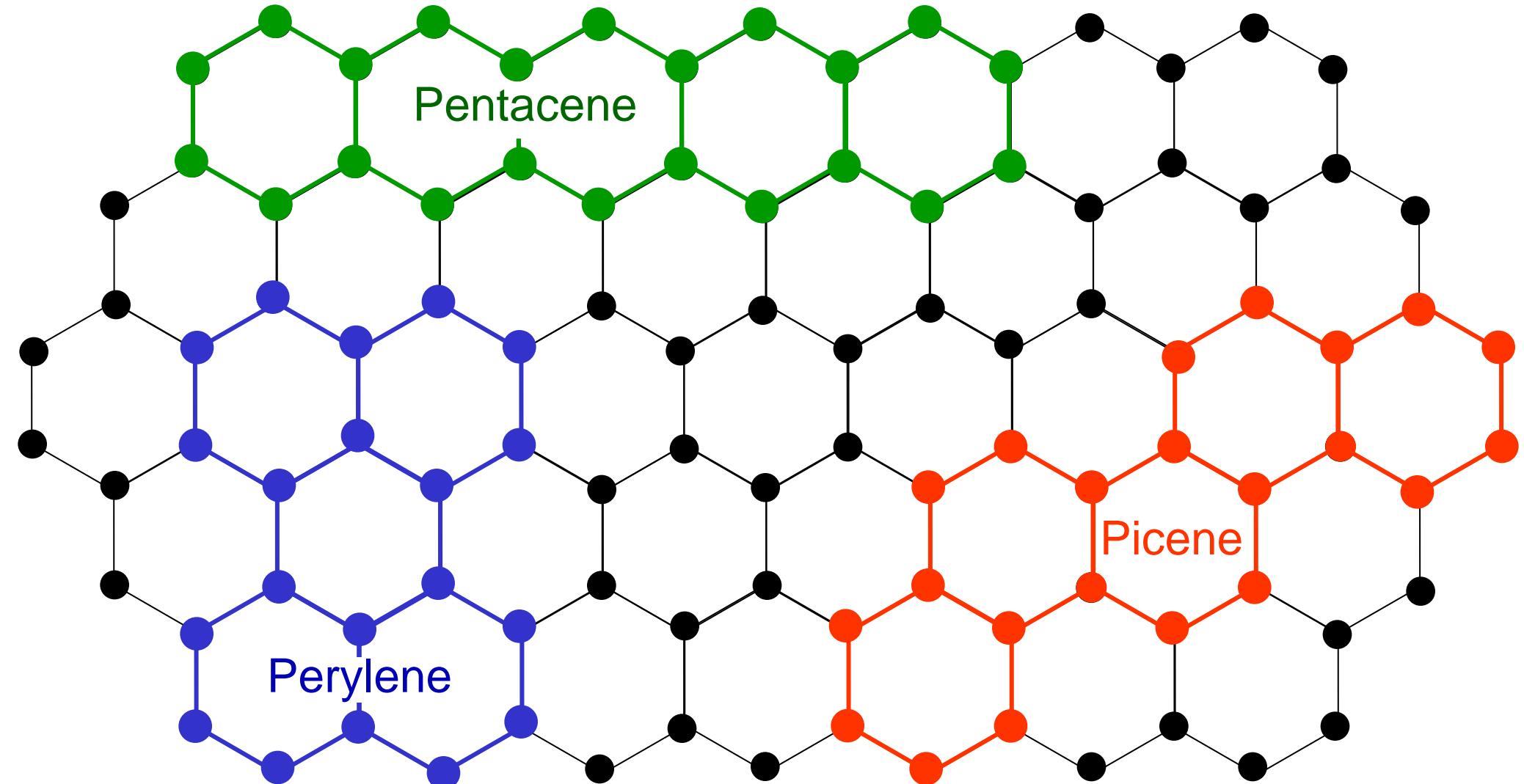


Tight-binding model: S. Katayama, A. Kobayashi, and Y. Suzumura, *J. Phys. Soc. Jpn.*, **75**, 054705 (2006).

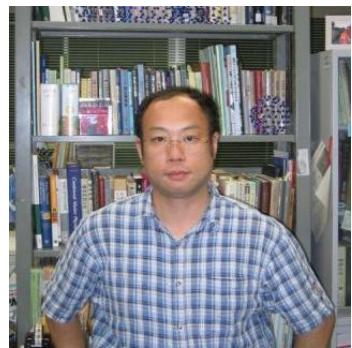
First principle band calculations : H. Kino and T. Miyazaki, *J. Phys. Soc. Jpn.*, **75**, 034704 (2006).

Fragments of graphene

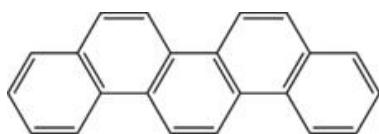
32



Superconductivity in potassium-doped picene 33



Prof. Y. Kubozono
(Okayama Univ.)



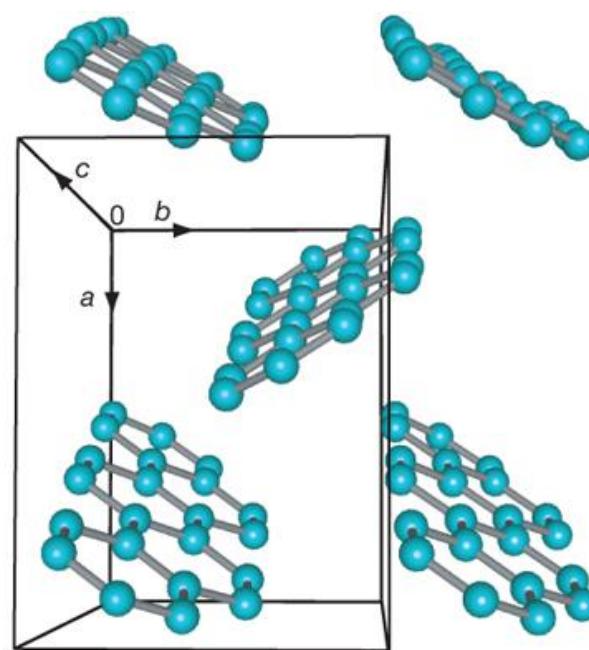
Picene



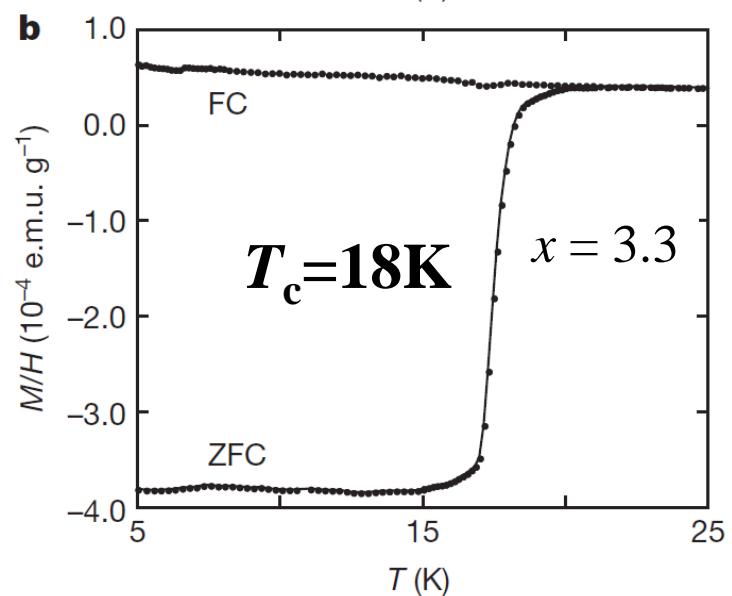
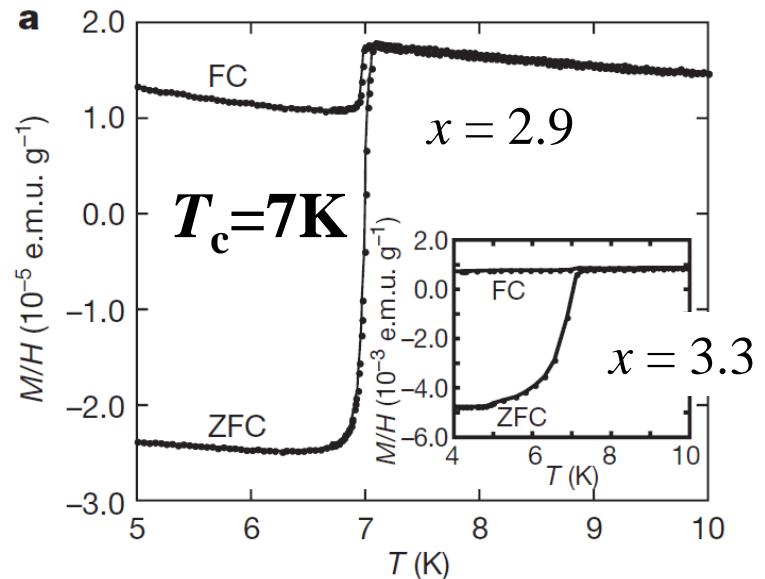
Picene crystals



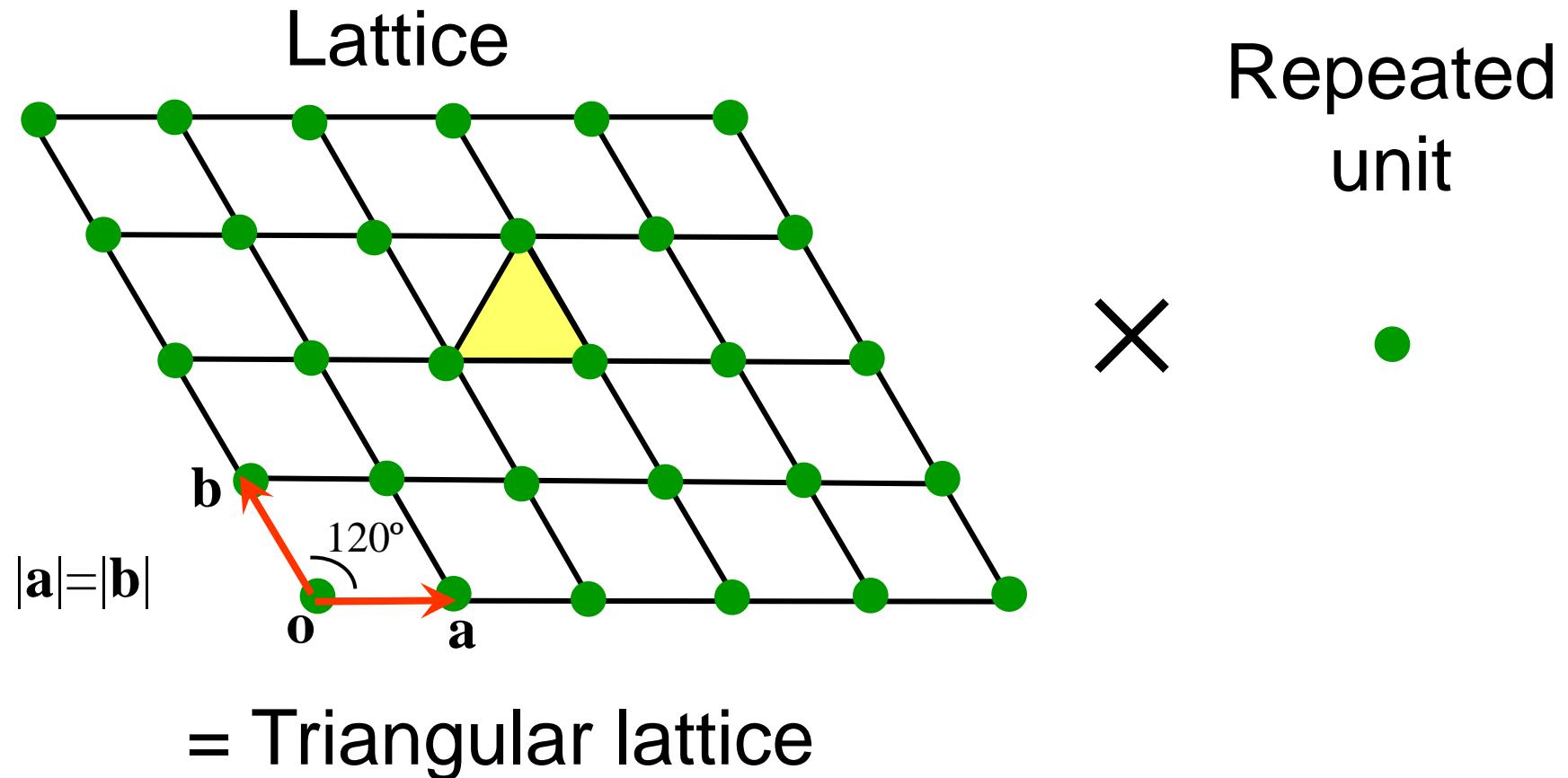
K_x picene



Nature, 464, 76 (2010)

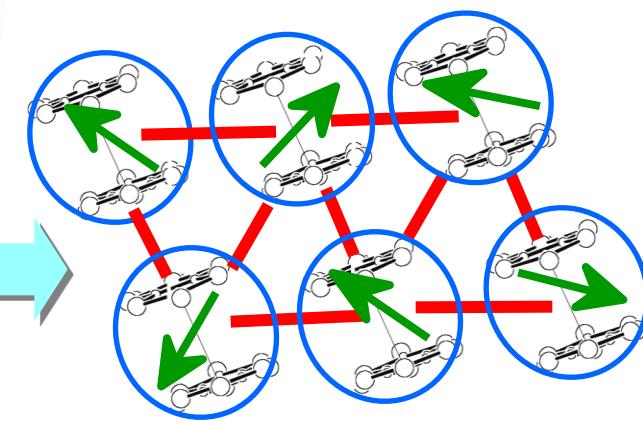
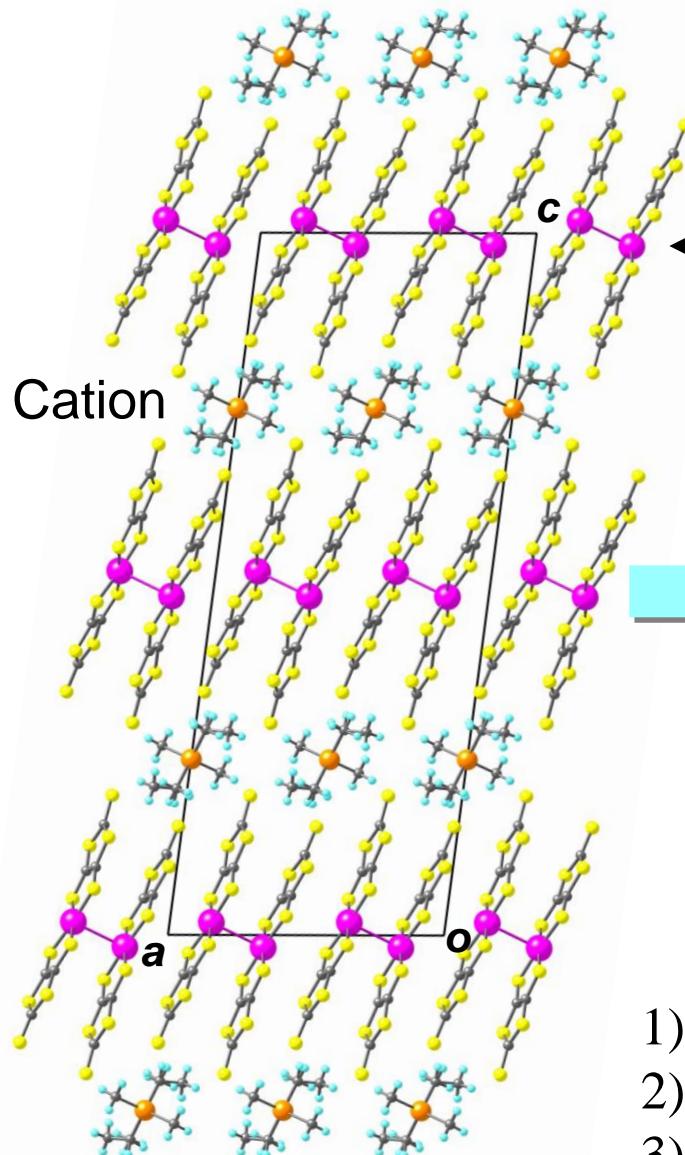


Triangular lattice

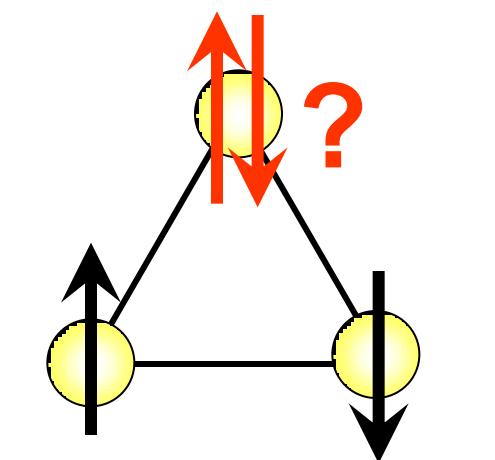
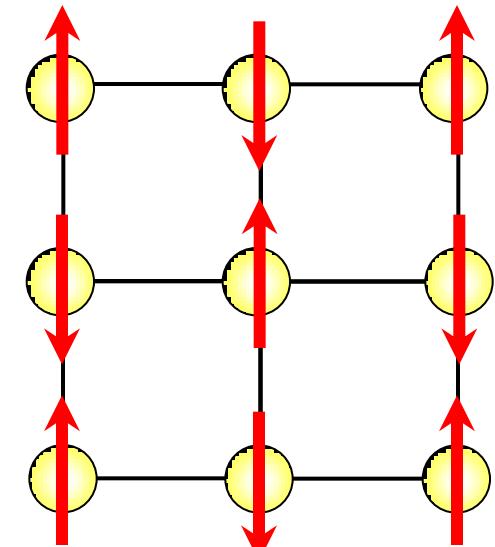


$$E(\mathbf{k}) = \beta(2\cos\mathbf{k}\cdot\mathbf{a} + 2\cos\mathbf{k}\cdot\mathbf{b} + 2\cos\mathbf{k}\cdot(\mathbf{a}+\mathbf{b}))$$

Spin liquid: EtMe₃Sb[Pd(dmit)₂]₂



- 1) *Science*, **328**, 1246 (2010).
- 2) *Nature Physics*, **6**, 673 (2010).
- 3) *Chem. Rev.*, **104**, 5319 (2004).



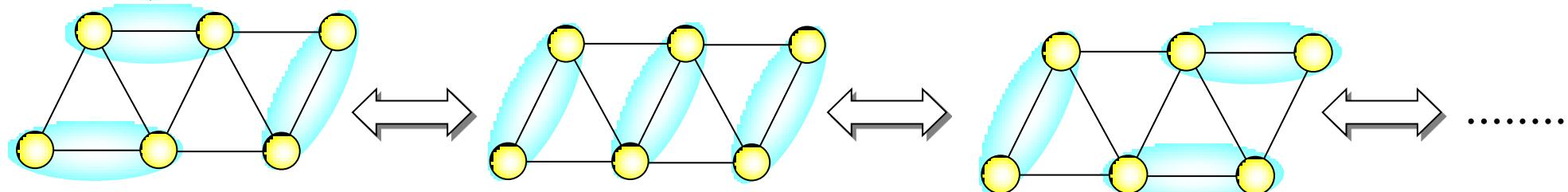
Spin frustration

Quantum Spin Liquid (QSL)

- QSL: A spin system in which quantum fluctuations prevent order, leading to liquid-like properties among the spins, even at zero temperature.

Singlet spin pair
(Valence Bond)

Resonation between highly degenerated spin configurations leads to a liquid-like wavefunction.



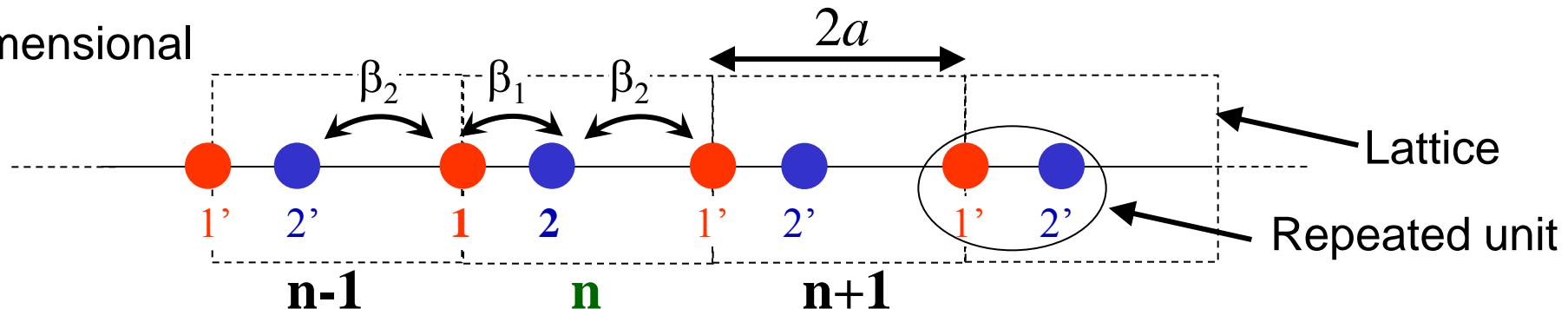
- A long-sought state of matter that has attracted much theoretical attention since its proposal by P. W. Anderson in 1973.
- There are few candidates of the real materials.
(Two in the molecular system: κ -(ET)₂Cu₂(CN)₃, EtMe₃Sb[Pd(dmit)₂]₂)

Summary

1. The electron is the lead of solid state science.
2. The state of the electrons can be represented as a linear combination of a set of base states with suitable coefficients (probability amplitudes).
3. The energy states of the electron can be obtained by the Hamiltonian matrix.
4. Metals are materials that have partially-filled energy bands.
5. Crystals are described by a (Bravais) lattice and a repeated unit.
6. Electronic properties strongly depend on crystal structures.

Appendix: Dimerized chain (I)

One-dimensional
crystal



$$\begin{cases} i\hbar \frac{dC_{1,n}}{dt} = \alpha C_{1,n} + \beta_1 C_{2,n} + \beta_2 C_{2,n-1} \\ i\hbar \frac{dC_{2,n}}{dt} = \alpha C_{2,n} + \beta_1 C_{1,n} + \beta_2 C_{1,n+1} \end{cases}$$

In the stationary state, $C_{i,n}(t) = A_i(n) \cdot e^{-\frac{i}{\hbar}Et}$

$$\begin{cases} EA_1(n) = \alpha A_1(n) + \beta_1 A_2(n) + \beta_2 A_2(n-1) \\ EA_2(n) = \alpha A_2(n) + \beta_1 A_1(n) + \beta_2 A_1(n+1) \end{cases}$$

Dimerized chain (II)

Let's take as a trial solution $A_i(n) = A_i e^{ik \cdot 2an}$

$$\begin{cases} (\alpha - E)A_1 + (\beta_1 + \beta_2 e^{-ik \cdot 2a})A_2 = 0 \\ (\beta_1 + \beta_2 e^{ik \cdot 2a})A_1 + (\alpha - E)A_2 = 0 \end{cases}$$

This is a set of linear algebraic equations for the unknowns A_1 and A_2 , and there is a solution if the determinant of the coefficients of A_1 and A_2 is zero.

$$\begin{vmatrix} \alpha - E & \beta_1 + \beta_2 e^{-ik \cdot 2a} \\ \beta_1 + \beta_2 e^{ik \cdot 2a} & \alpha - E \end{vmatrix} = 0$$

$$\therefore E(k) = \alpha \pm \sqrt{\beta_1^2 + \beta_2^2 + 2\beta_1\beta_2 \cdot \cos 2ka}$$

Energy band for a dimerized chain A3

Energy band for
a uniform chain

